



How Wine/Cider Components Affect Shelf Life in Cans

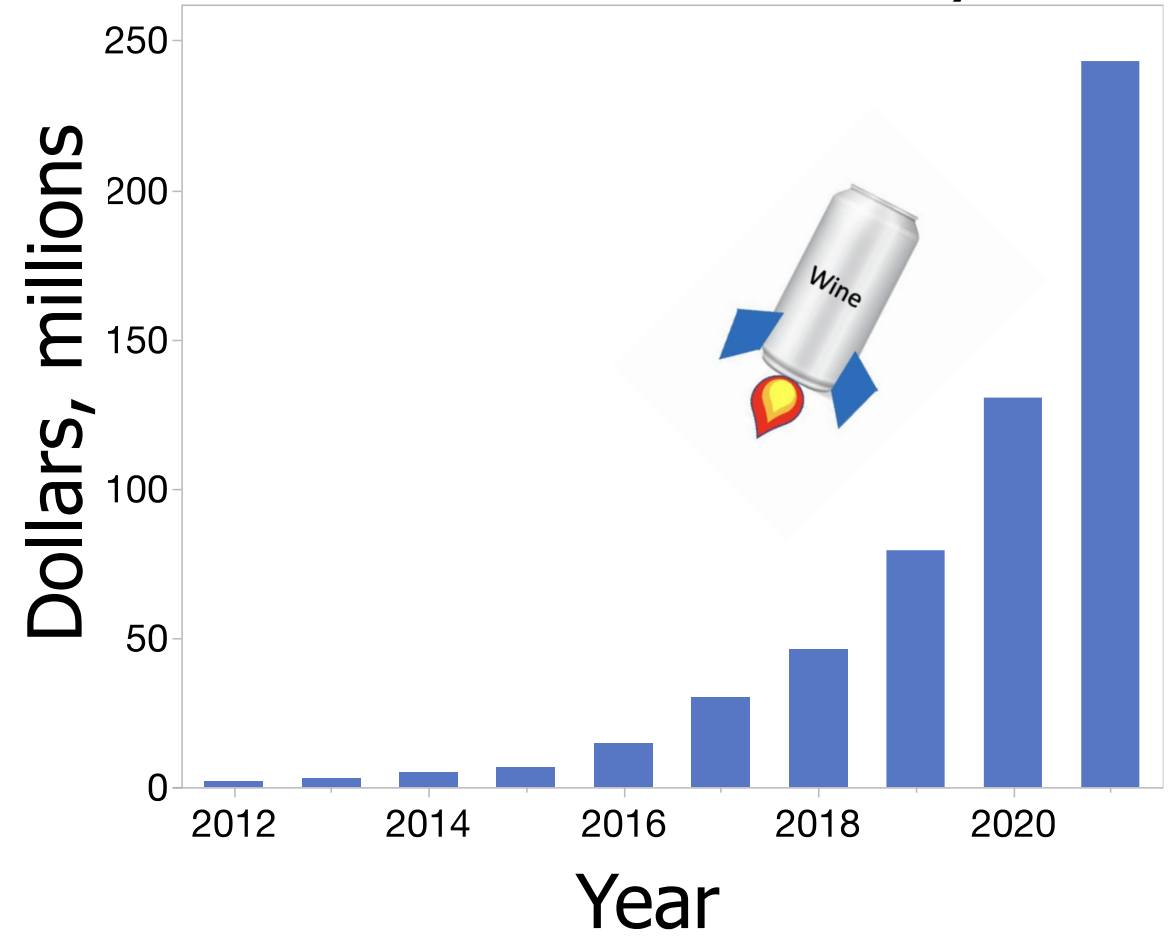
Congrès cidres, vins et alcools d'ici 2024

Austin Montgomery

Canned wine – trendy packaging

- Fastest growing packaging type

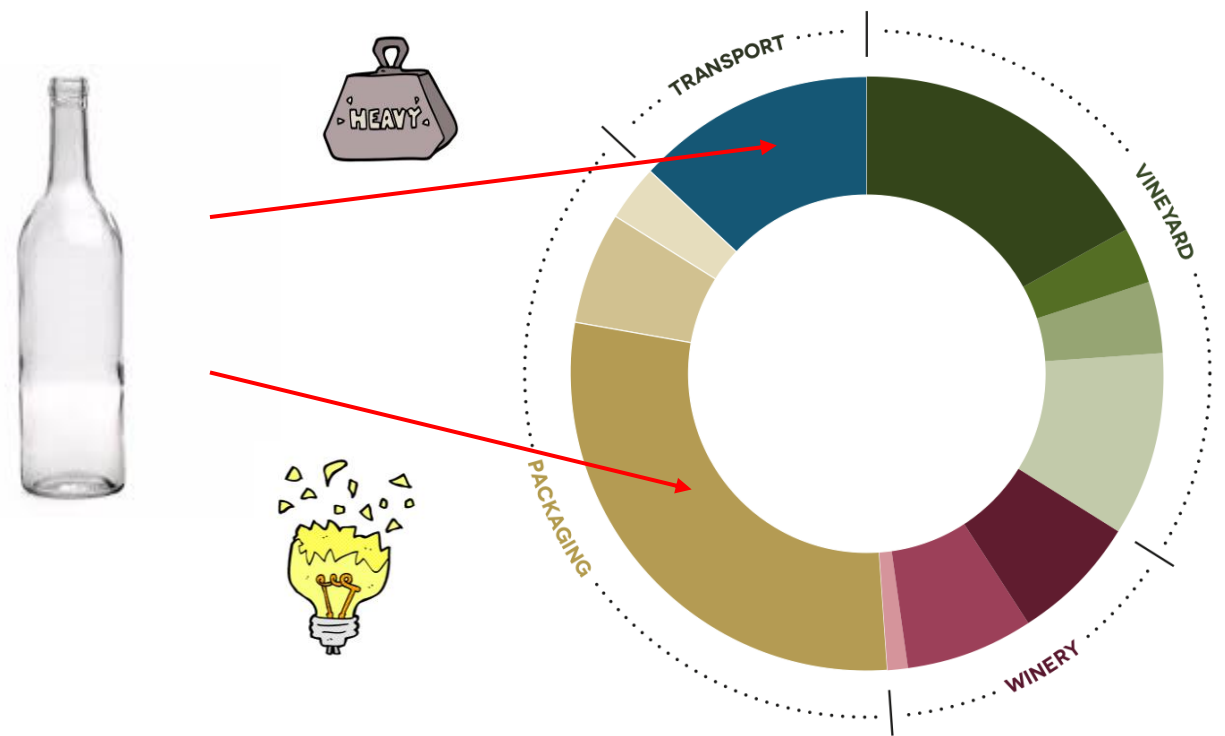
US Canned Wines Profit by Year



Canned wine – trendy packaging

- Fastest growing packaging type
- Less carbon footprint

Relative impact for carbon footprint of wines



Canned wine – trendy packaging

- Fastest growing packaging type
- Less carbon footprint
- Convenient Packaging



Can Specific Problems

Problem	Description
Staling	<i>Accelerated oxidation (if air in headspace; or, following corrosion and loss of hermetic seal)</i>



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Scalping	<i>Absorption of odorants by polymeric liner (possibly an issue for hop aroma/cannabis compounds)</i>



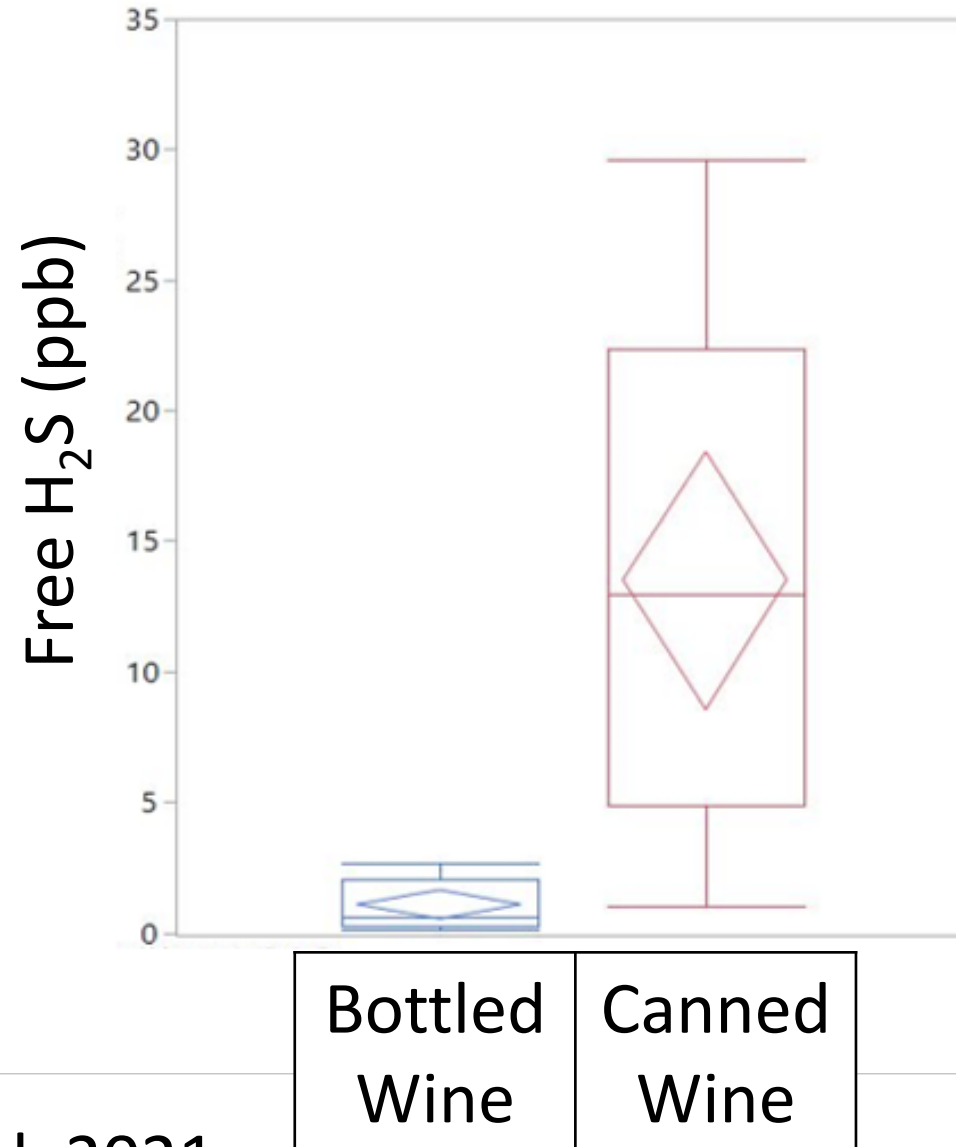
Can Specific Problems

Problem	Description
Staling	<i>Accelerated oxidation (if air in headspace; or, following corrosion and loss of hermetic seal)</i>
Scalping	<i>Absorption of odorants by polymeric liner (possibly an issue for hop aroma/cannabis compounds)</i>
Tainting	<i>Haze or off-flavor formation from aluminum or other components following corrosion</i>



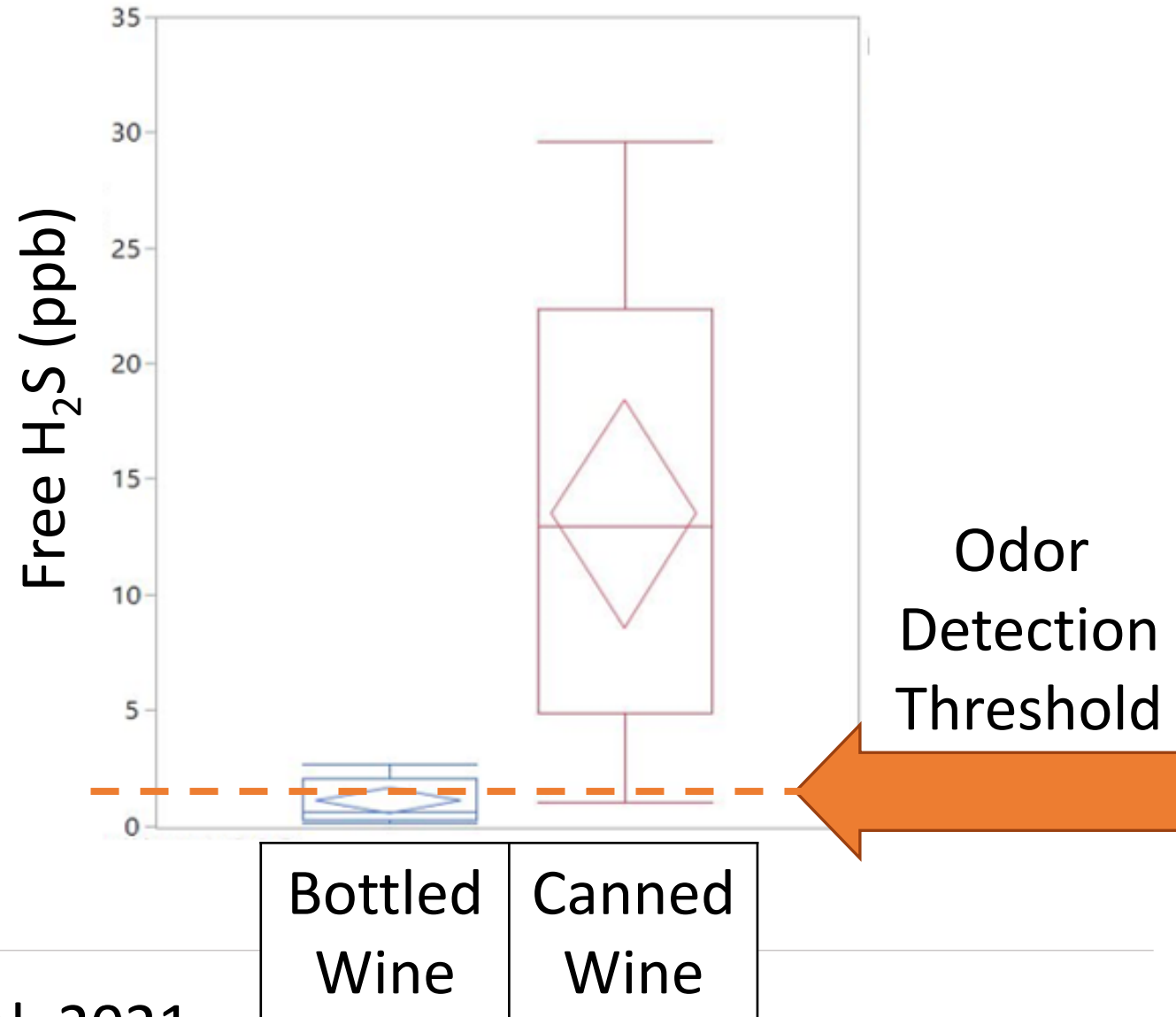
Canned wines typically have higher H₂S than glass

- Average H₂S of canned wines (~13 ppb)



Canned wines typically have higher H₂S than glass

- Average H₂S of canned wines (~13 ppb)
- H₂S odor detection threshold: 1.6 ppb (Siebert et al.)

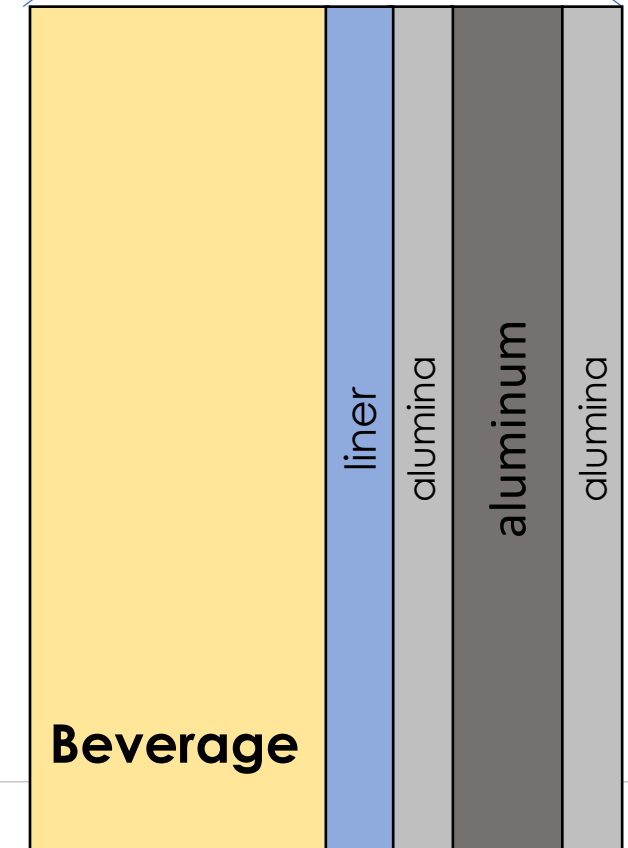


Anatomy of a can

- A can is basically a thin plastic bottle with an aluminum shell.
- Typical liners we have studied:
 - BPA-epoxy
 - BPA-NI alternatives
 - Acrylic
 - Polyester
 - BPF epoxy
 - aTULC



zoom in



Anatomy of a can

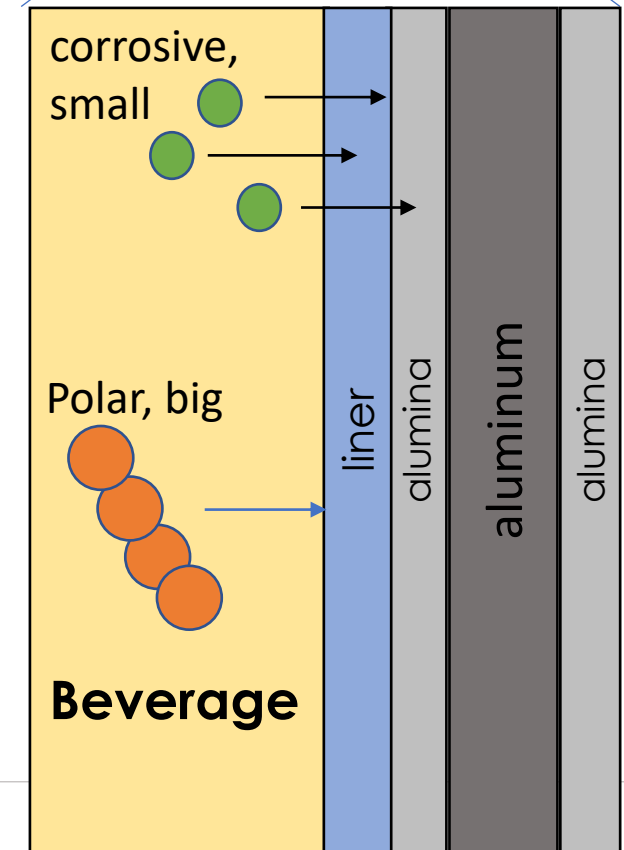
- A can is basically a thin plastic bottle with an aluminum shell.
- Two places to focus on corrosion:
 - 1) **Liner x beverage interactions**
 - 2) Aluminum x beverage interactions

“How fast does a compound diffuse through the polymer?”

“How do beverage components affect diffusion?”



zoom in



Anatomy of a can

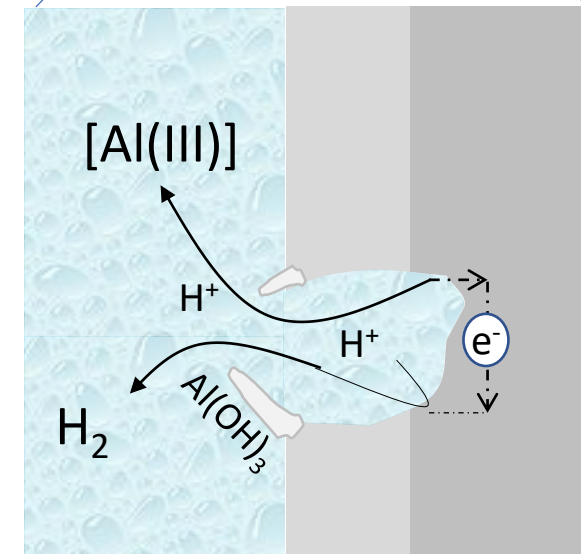
- A can is basically a thin plastic bottle with an aluminum shell.
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 - 2) Aluminum x beverage interactions**

“What reaction occurs between beverage components and aluminum?”

“What can we measure to quantify this reaction?”



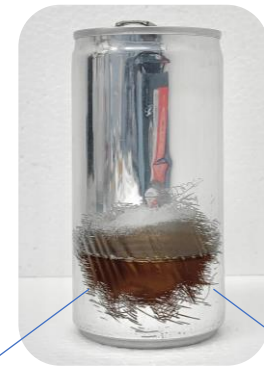
zoom in



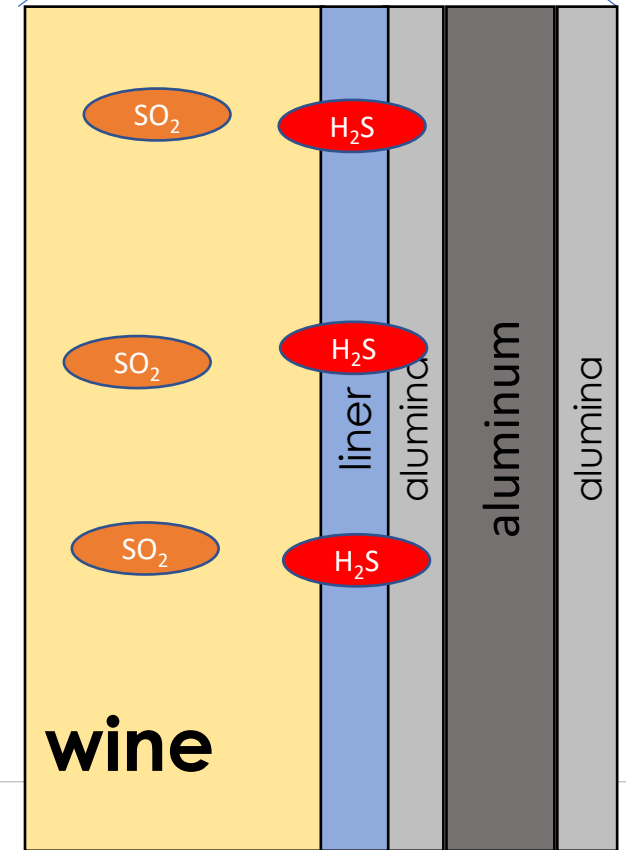
H₂S formation



Store @ 20 ° C
4-8 months



zoom in



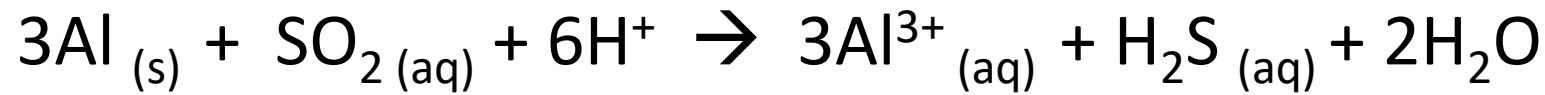
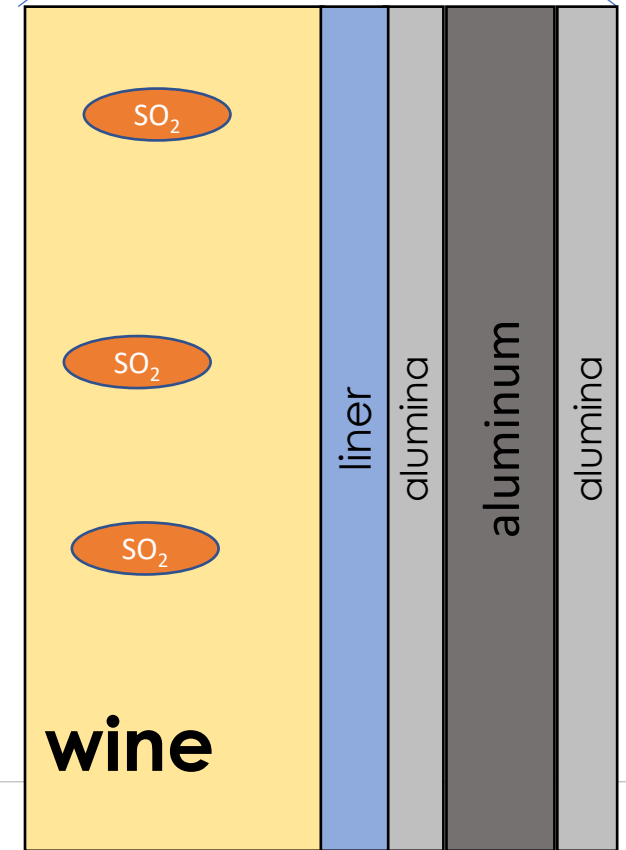
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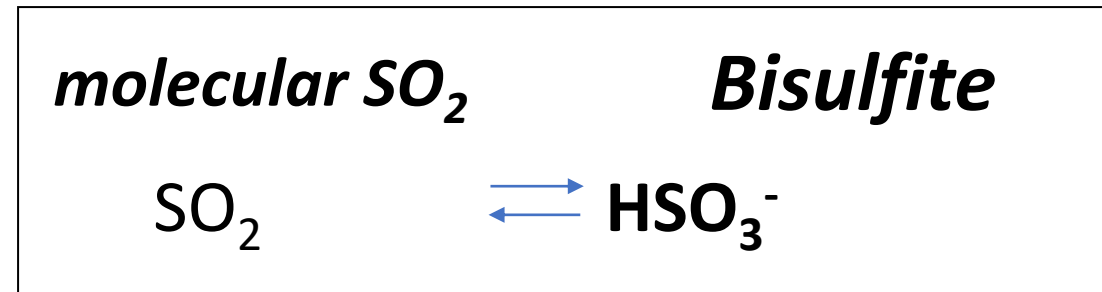
zoom in



Or, SO₂ facilitates reduction of other sources of sulphur

Review of SO_2 in wine: free vs. molecular

“Free SO_2 ” = molecular + bisulfite



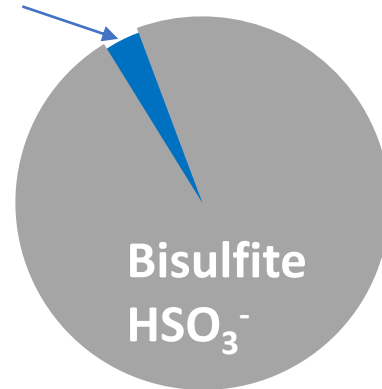
Ratio depends on pH

Molecular SO_2 is a minor component of free SO_2 , typically $<5\%$

Higher molecular SO_2 proportions at lower pH

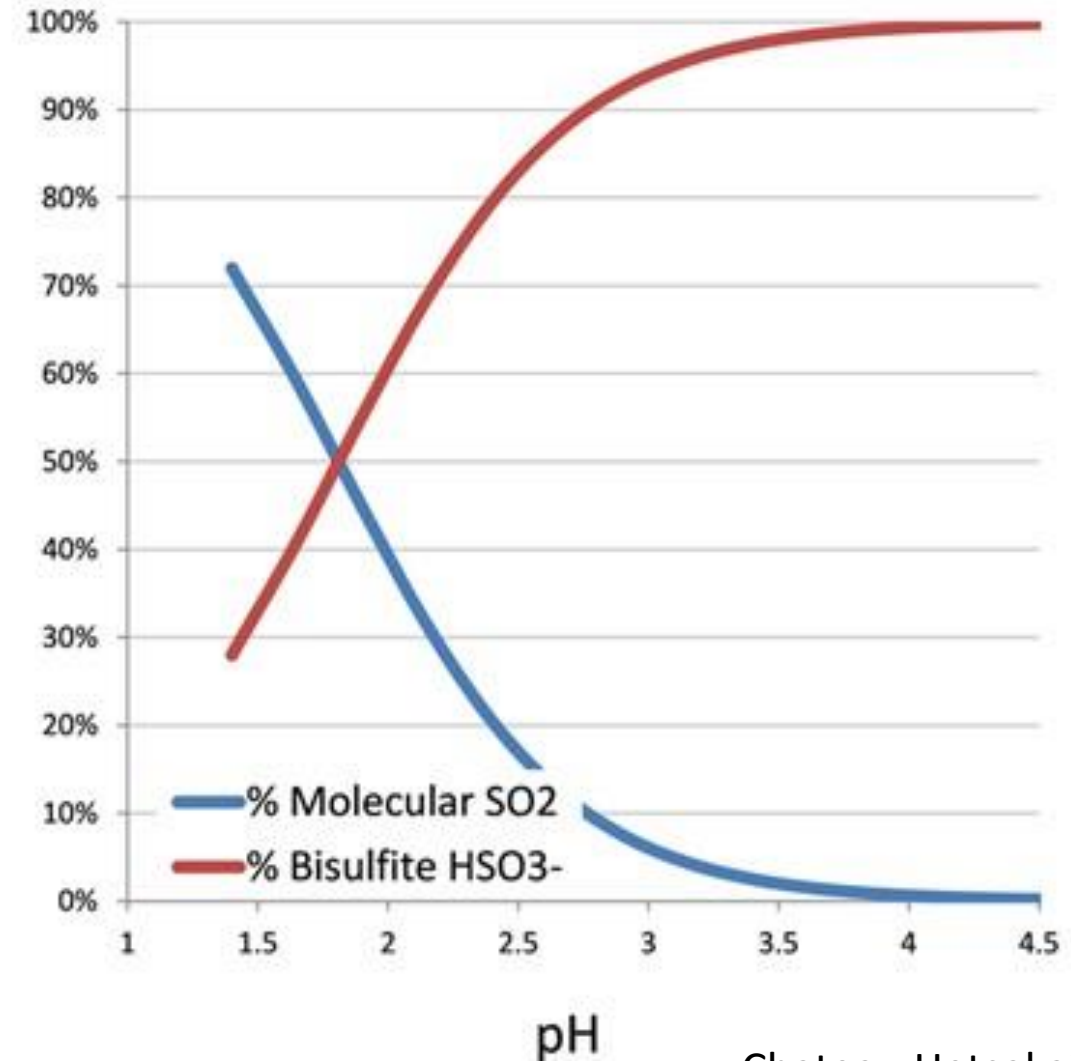
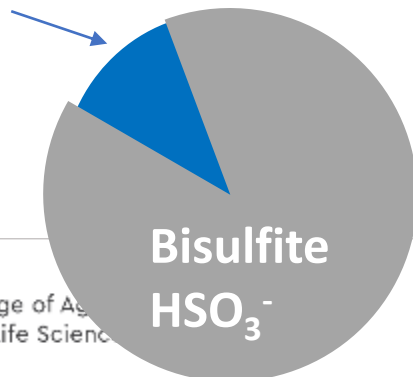
Lower proportion molecular SO_2

Higher pH = 3.4



Higher proportion molecular SO_2

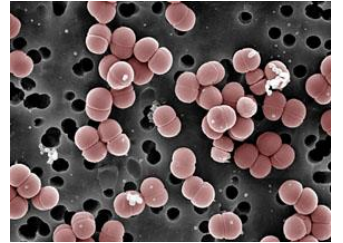
Lower pH = 2.8



Review: we have two complementary targets with SO₂ ¹⁶

1) Molecular SO₂ is an **antimicrobial**

Typically, **~0.6 mg/L molecular SO₂** recommended to prevent spoilage, with higher amounts recommended for wines with residual sugar
At higher concentrations, concerns about sensory effects



2) Free SO₂ (as bisulfite) is an **antioxidant**

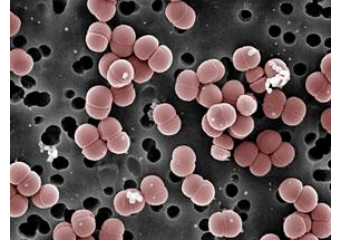
~30 mg/L **free SO₂** common recommendation for glass packaging
Reacts with oxidation products (e.g. hydrogen peroxide, quinones)
Typically, at **<10 mg/L free SO₂**, oxidized aromas appear



SO₂ should be limited in cans

1) Molecular SO₂ is an **antimicrobial**

-> **Currently not a better option for microbial stability at the same price point. In-Bulk HPP and velcorin are options**



2) Free SO₂ (as bisulfite) is an **antioxidant**

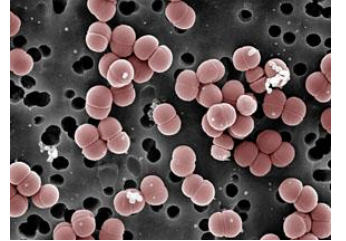
-> Cans hermetic seal eliminate the need for oxygen control in canned wines.



SO₂ should be limited in cans

1) Molecular SO₂ is an antimicrobial

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2) Free SO₂ (as bisulfite) is an **antioxidant**

-> Cans hermetic seal eliminate the need for oxygen control in **canned wines.**

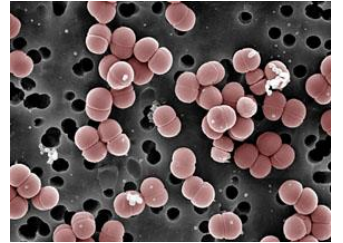


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What factors affect H₂S production?



?

Do any wine components predict H₂S production?



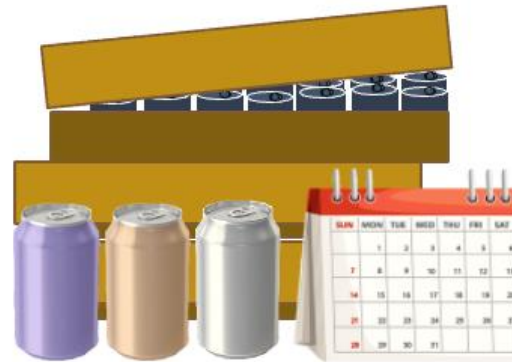
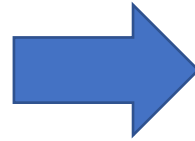
How does liner choice affect H₂S production following storage?

Experimental design



**10 wines
(2 rosés, 2
reds, 6 whites)**

Experimental design



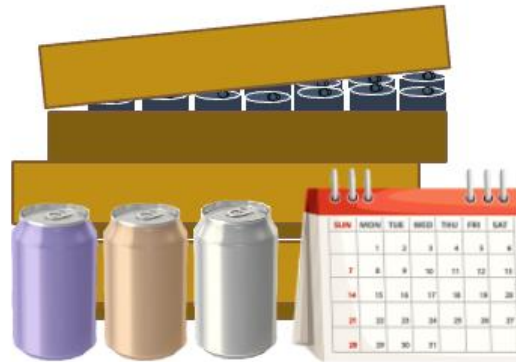
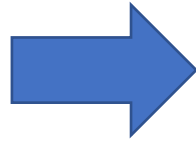
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**Canned with..
N = 3 Liners
N = 4 Timepoints
N = 3 Replicates**

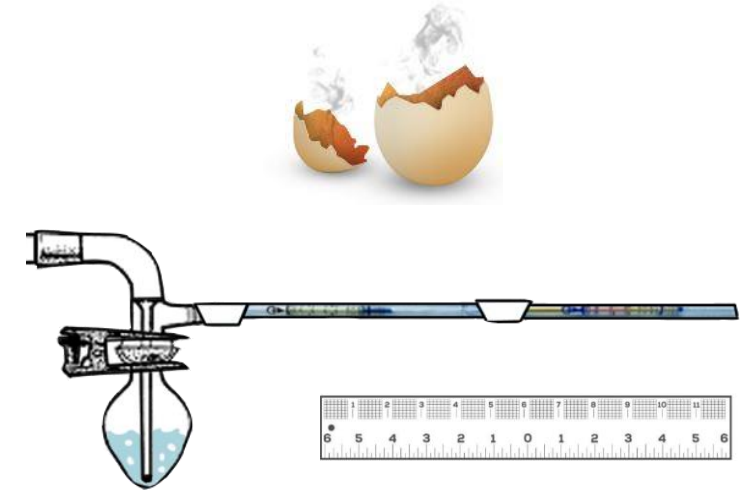
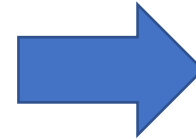
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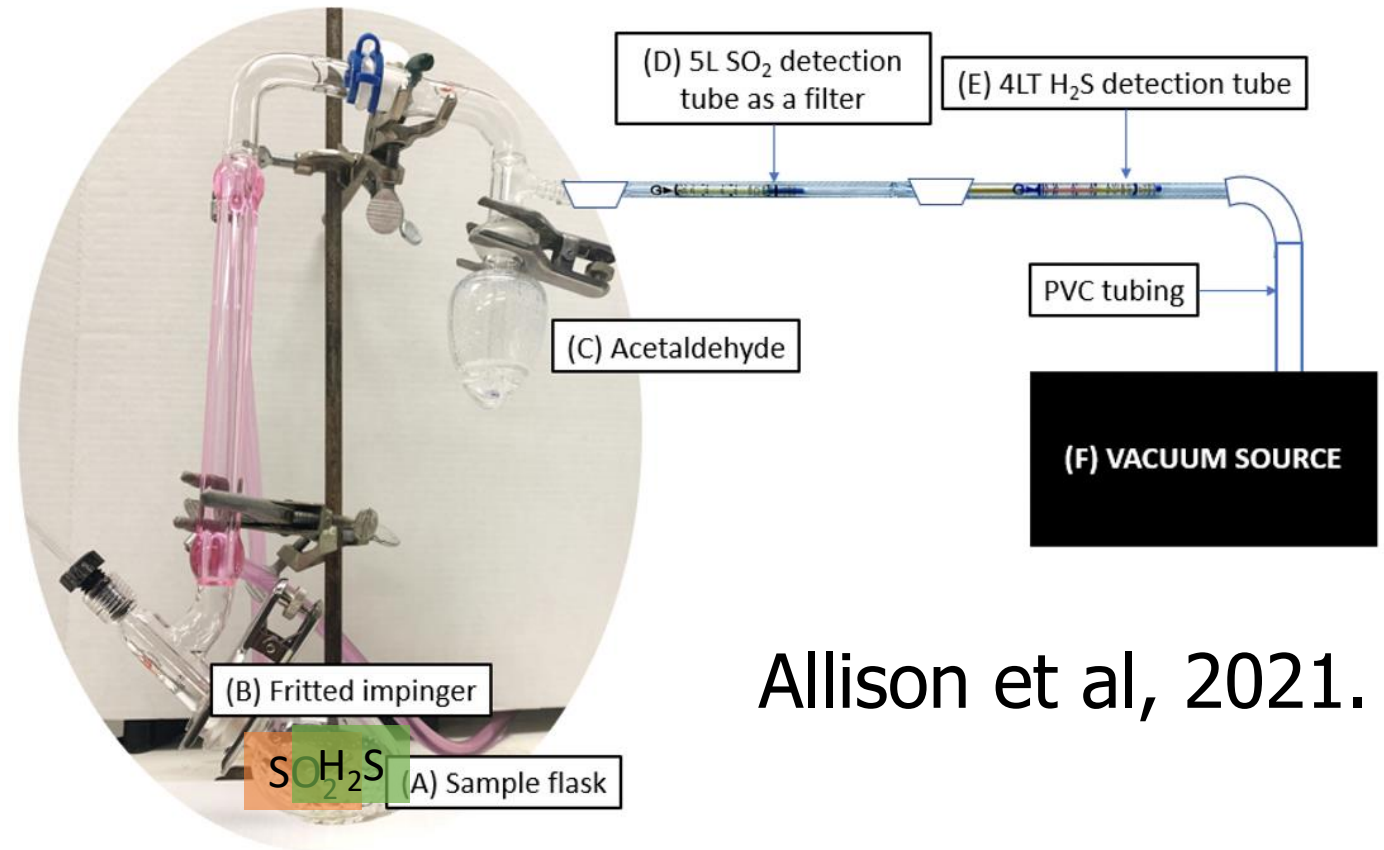
**Canned with..
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**Measure H₂S
Visible degradation
Measure Al³⁺**

Measuring H₂S with gas detection tubes

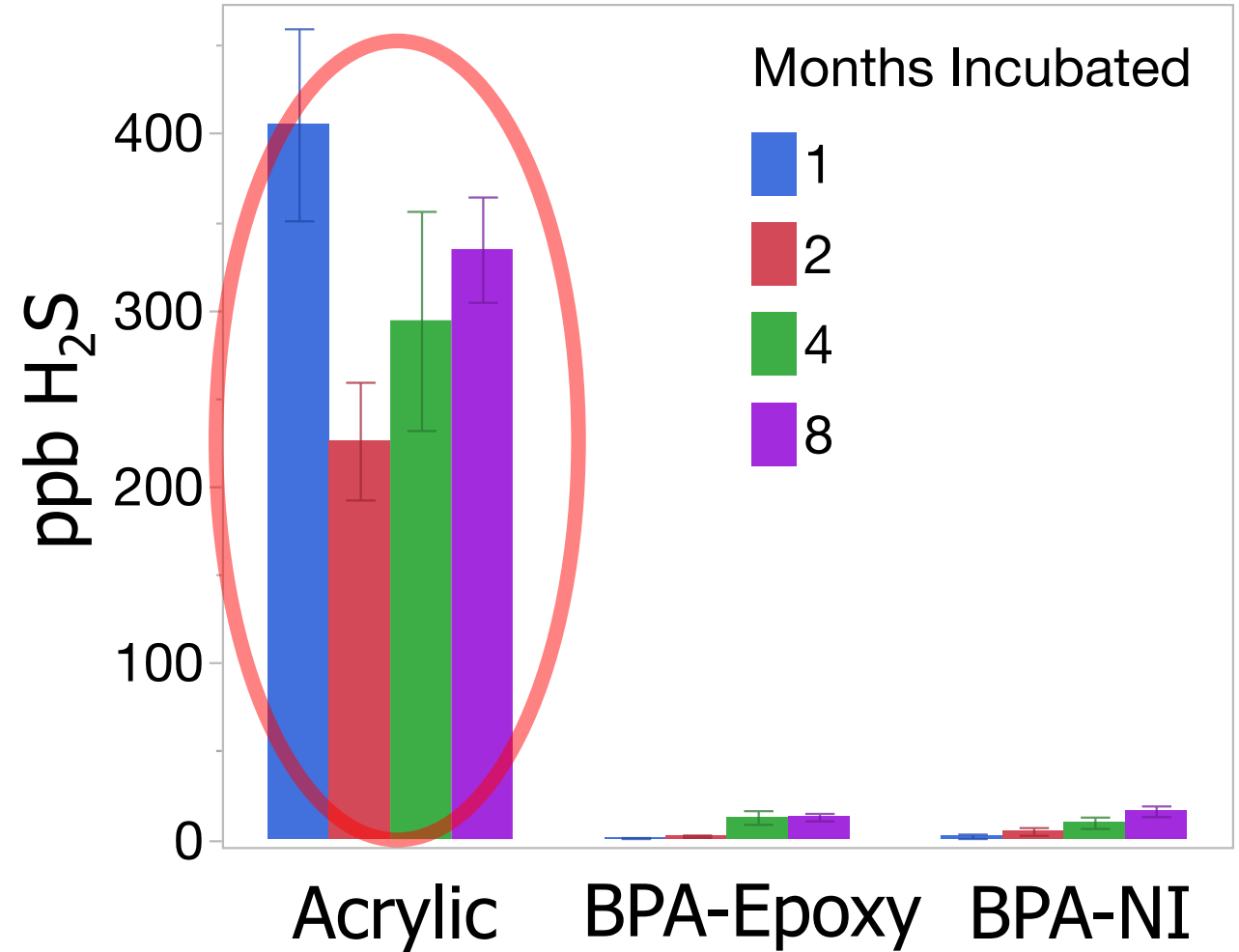
- Wineries can adapt A-O setup to measure H₂S <10 minutes
- Excellent reproducibility within standard solutions



Allison et al, 2021.

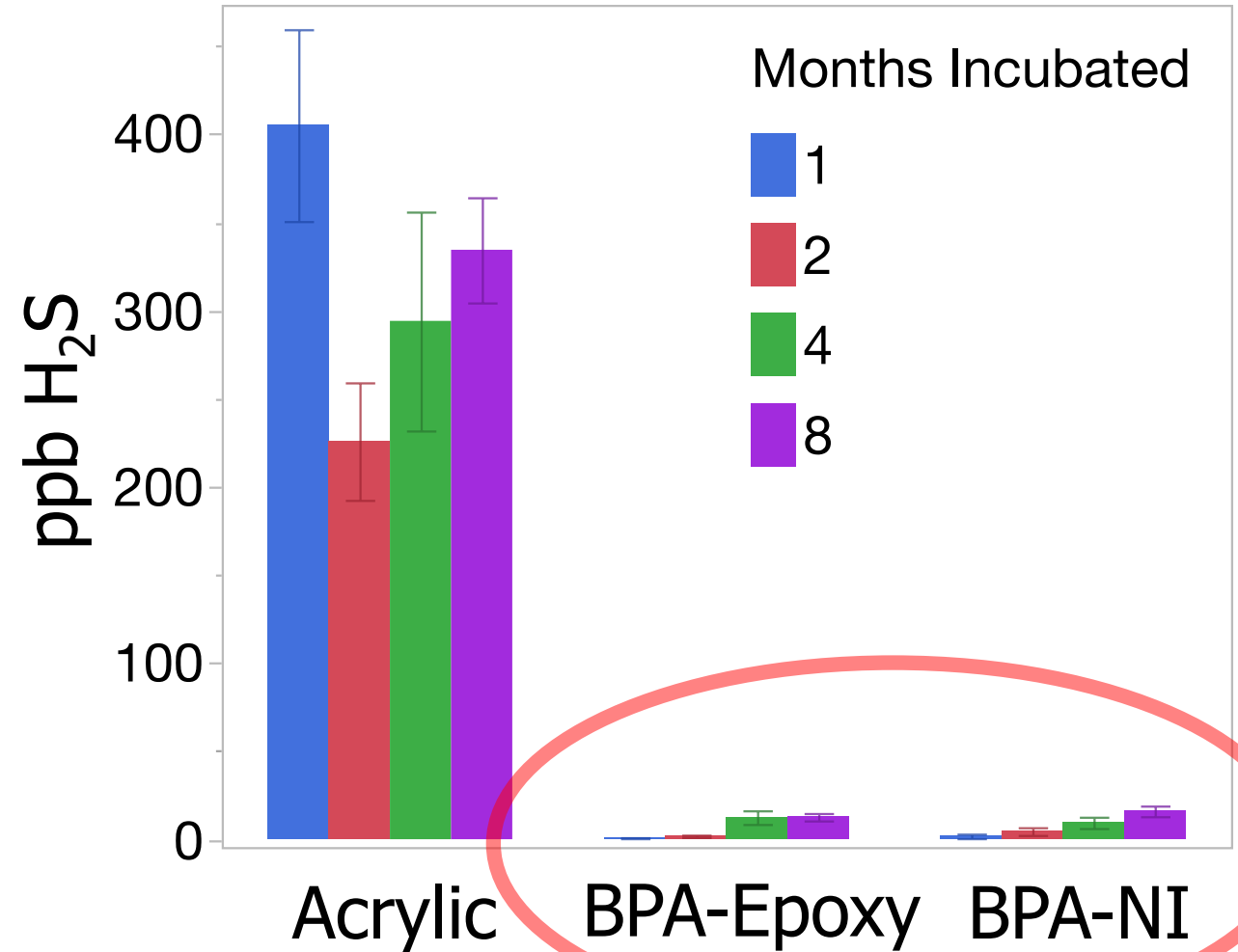
Effect of liner type – long term aging

- Acrylic is not a suitable liner for wine.



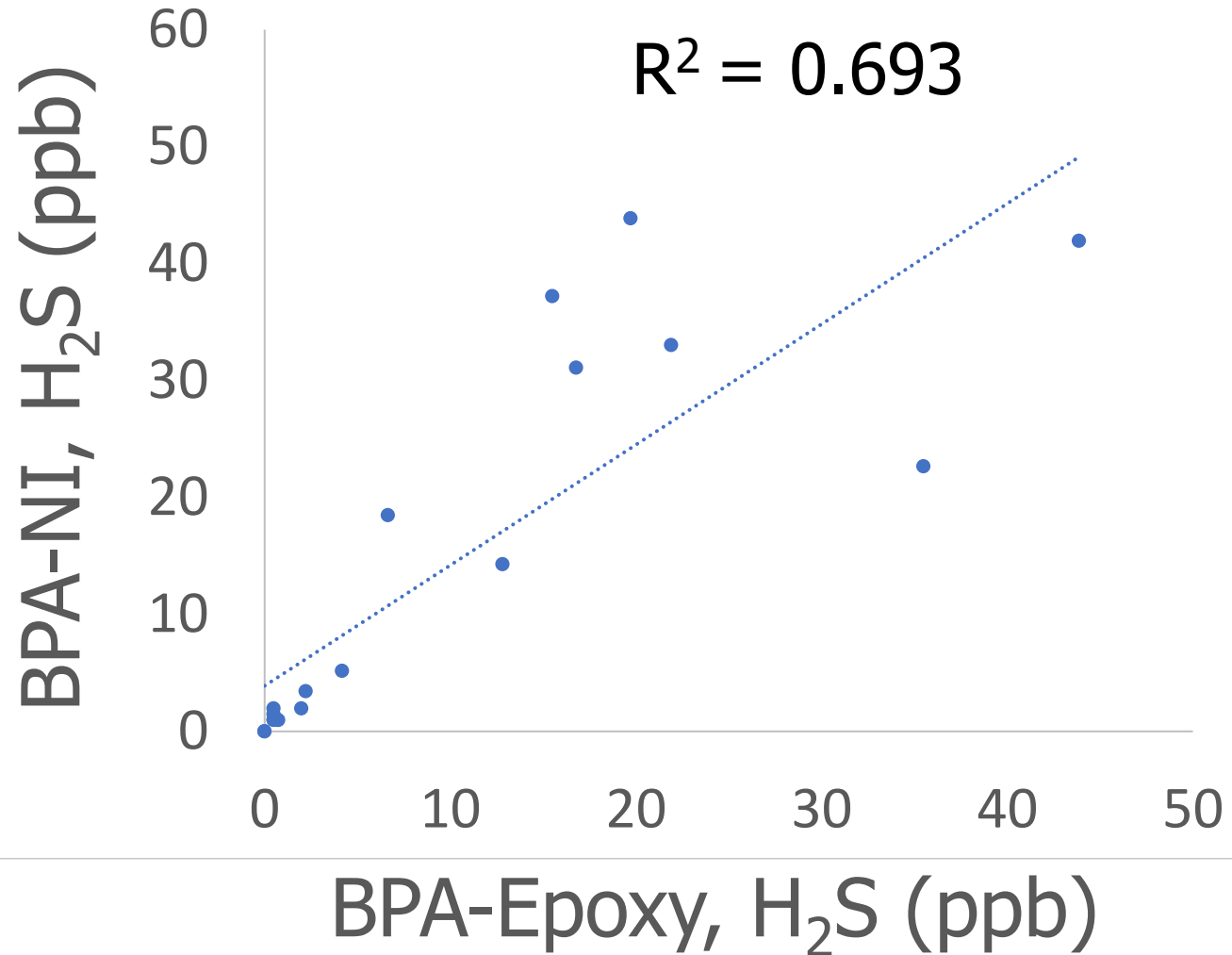
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- Acrylic is not a suitable liner for wine.
- BPA epoxy and BPA-NI perform comparably.

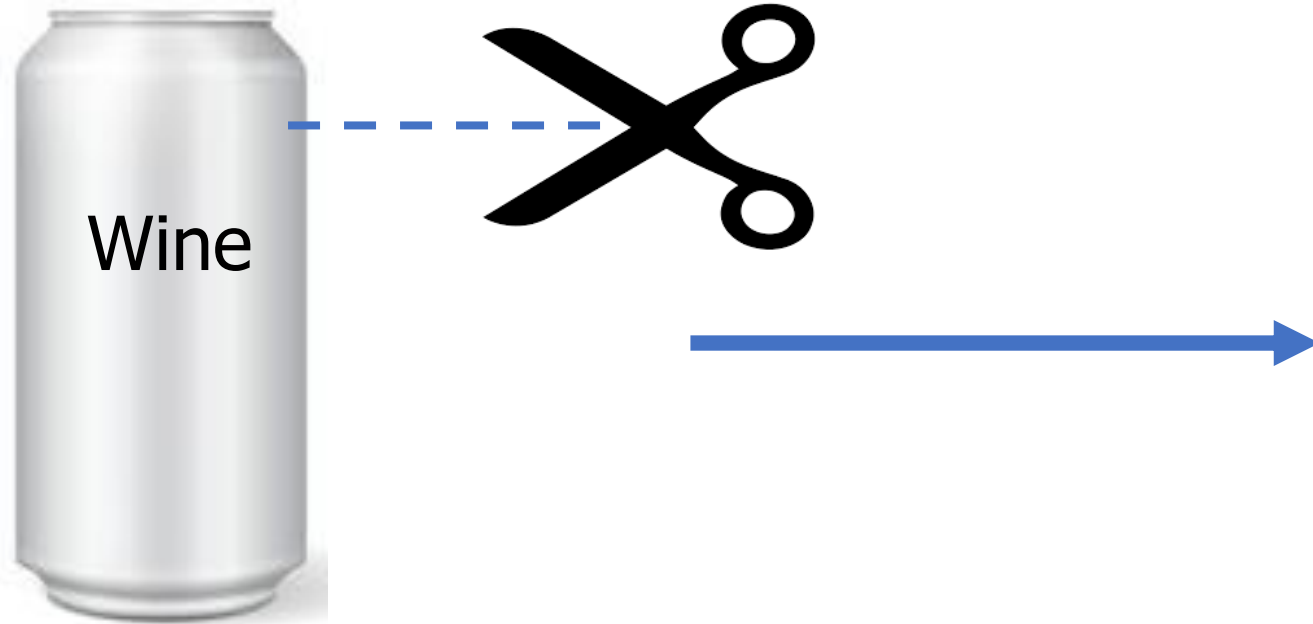


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Visible degradation correlates to H₂S production



BPA/BPA-NI

No H₂S Production

Visible degradation correlates to H₂S production

- Acrylic looked... scary.



Acrylic
High H₂S Producer



BPA/BPA-NI
No H₂S Production

Visible degradation correlates to H₂S production

- Acrylic looked... scary
- BPA/BPA-NI showed blistering in headspace + body damage



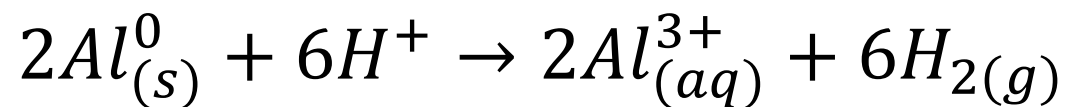
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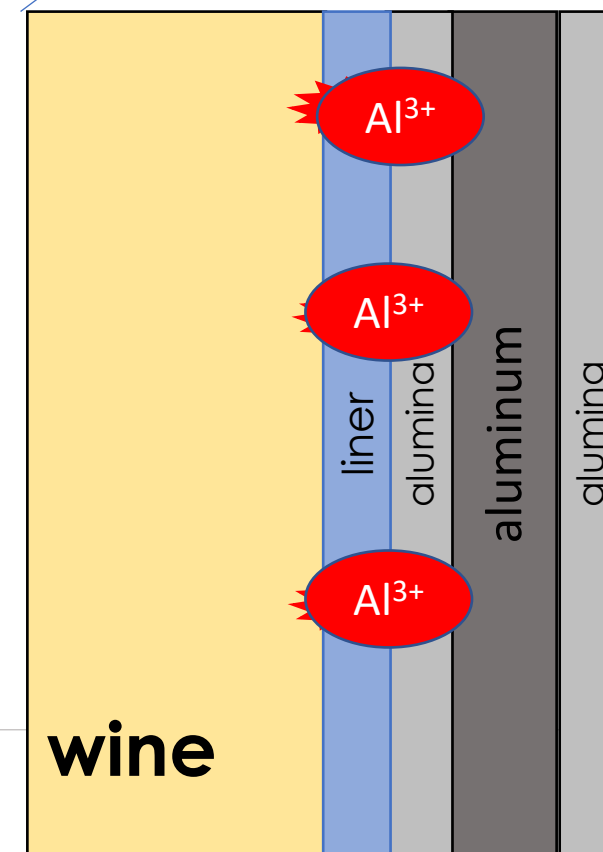
BPA/BPA-NI
No H₂S Production

Expectation: Visual degradation = dissolved Al^{3+}

- Visible liner damage should lead to the exposure of bare aluminum
- This should result in dissolved aluminum:

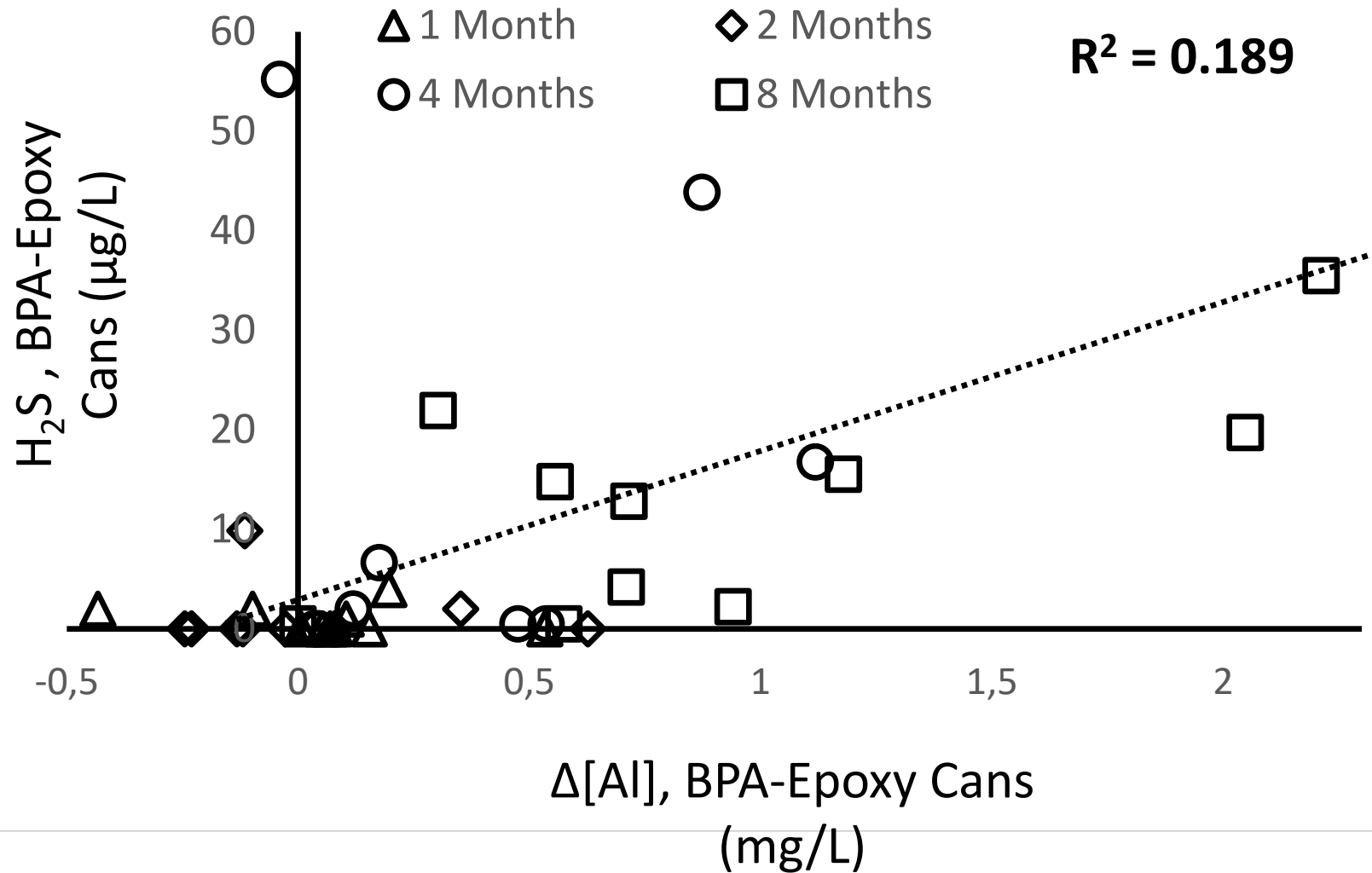


zoom in



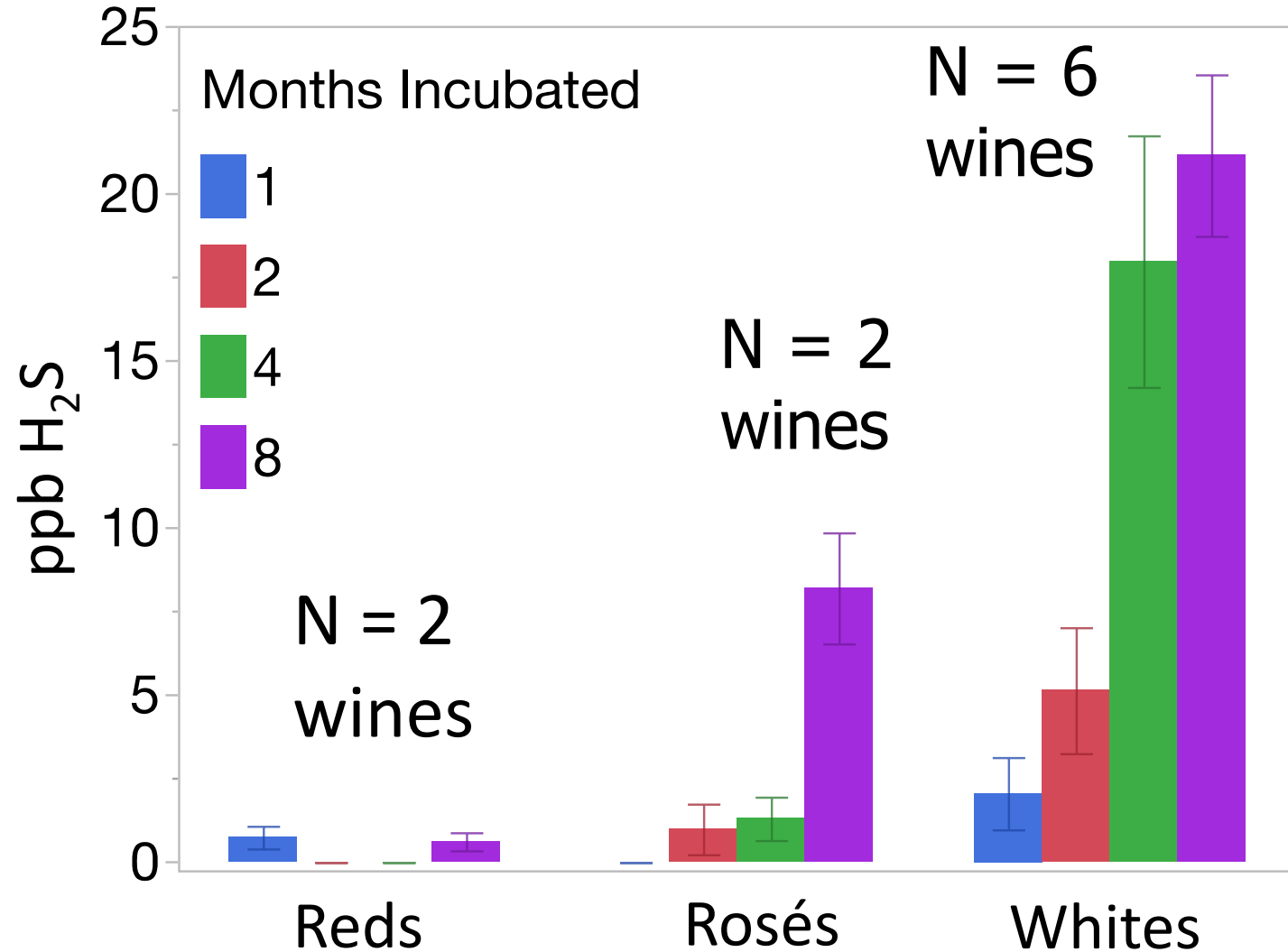
... but Al^{3+} is poorly correlated to H_2S production

- Negligible Al^{3+} signal until somewhere between 4 and 8 months.



Effect of wine type – long term aging

- Significant H₂S formation can happen as soon as 4 months



Chemical composition of wines studied

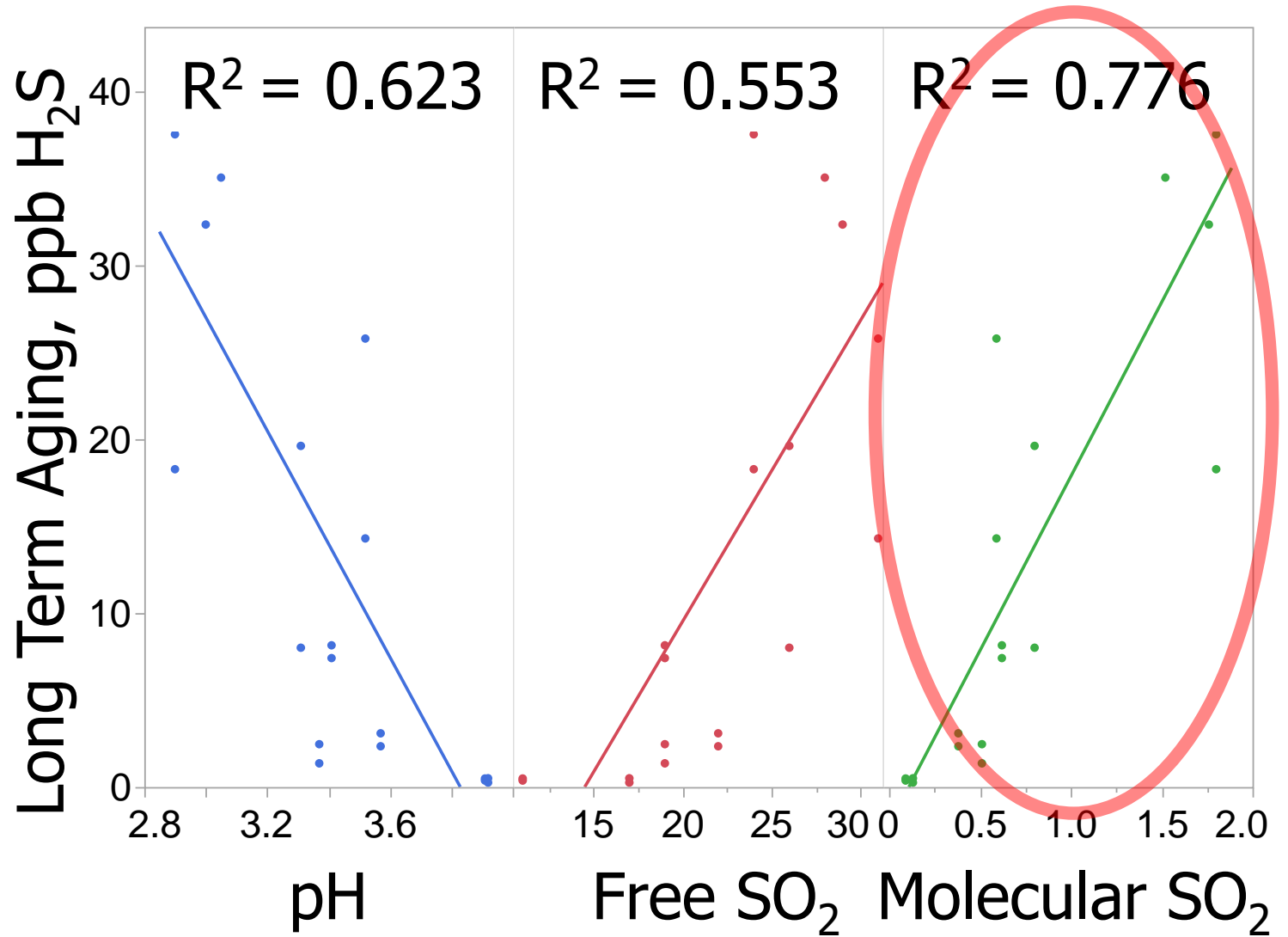
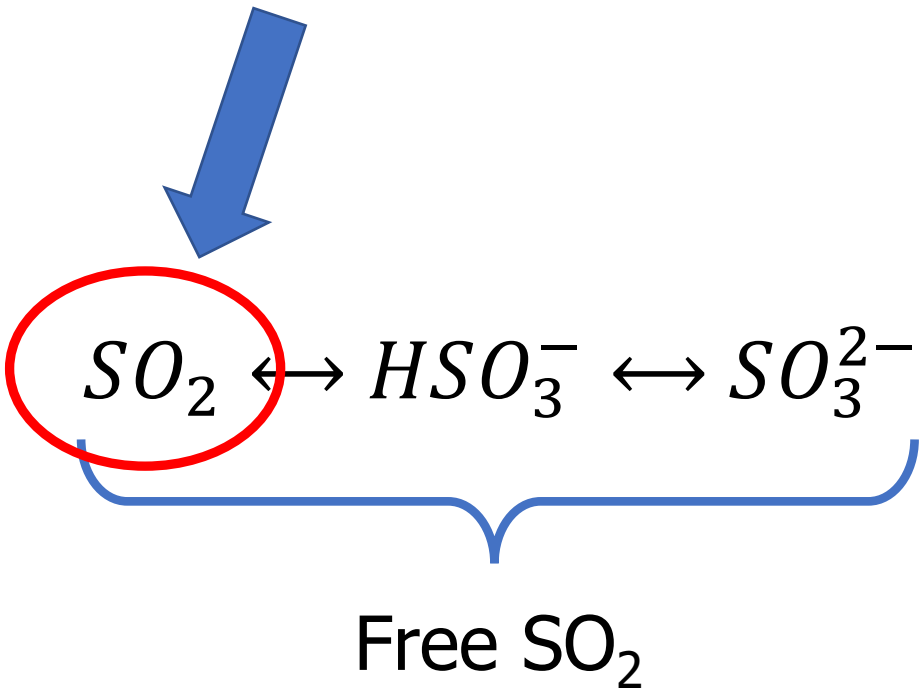
- Wines possessed typical table wine chemistry with these ranges:

Range	pH	TA g/L	ABV v/v	SO ₂ mg/L			Sugar g/L	Cu mg/L	Cl- mg/L
				Free	Total	Mol.			
Low	2.9	5.5	8.6	11	36	0.1	0.9	0.05	32.1
High	3.9	9.1	13.2	31	142	1.8	65.33	0.39	442

What best predicts H₂S?

Molecular SO₂

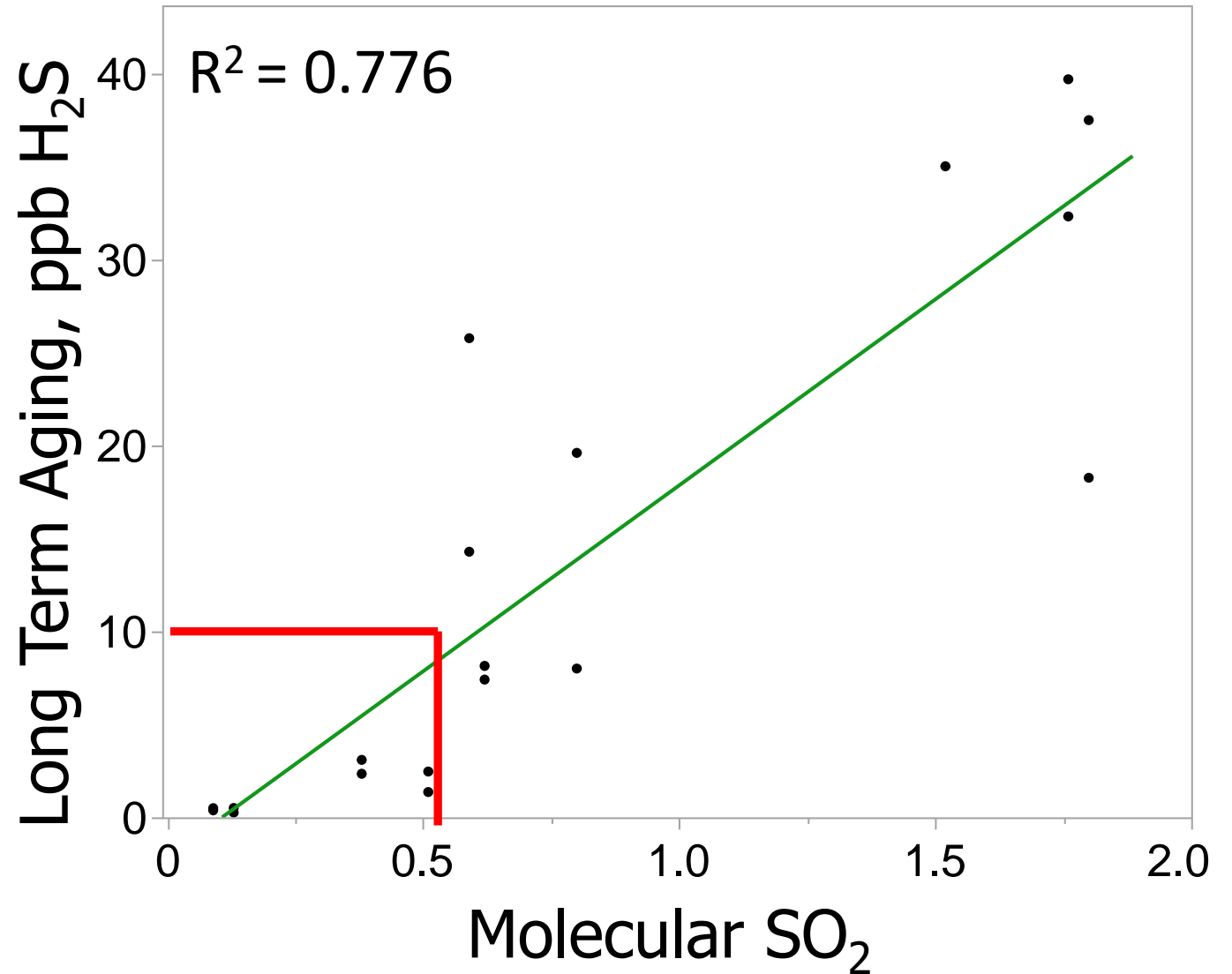
- Molecular SO₂ = best predictor of H₂S



What best predicts H₂S?

Molecular SO₂

- Molecular SO₂ = best predictor of H₂S
- ~ 0.5 ppm target Molecular SO₂



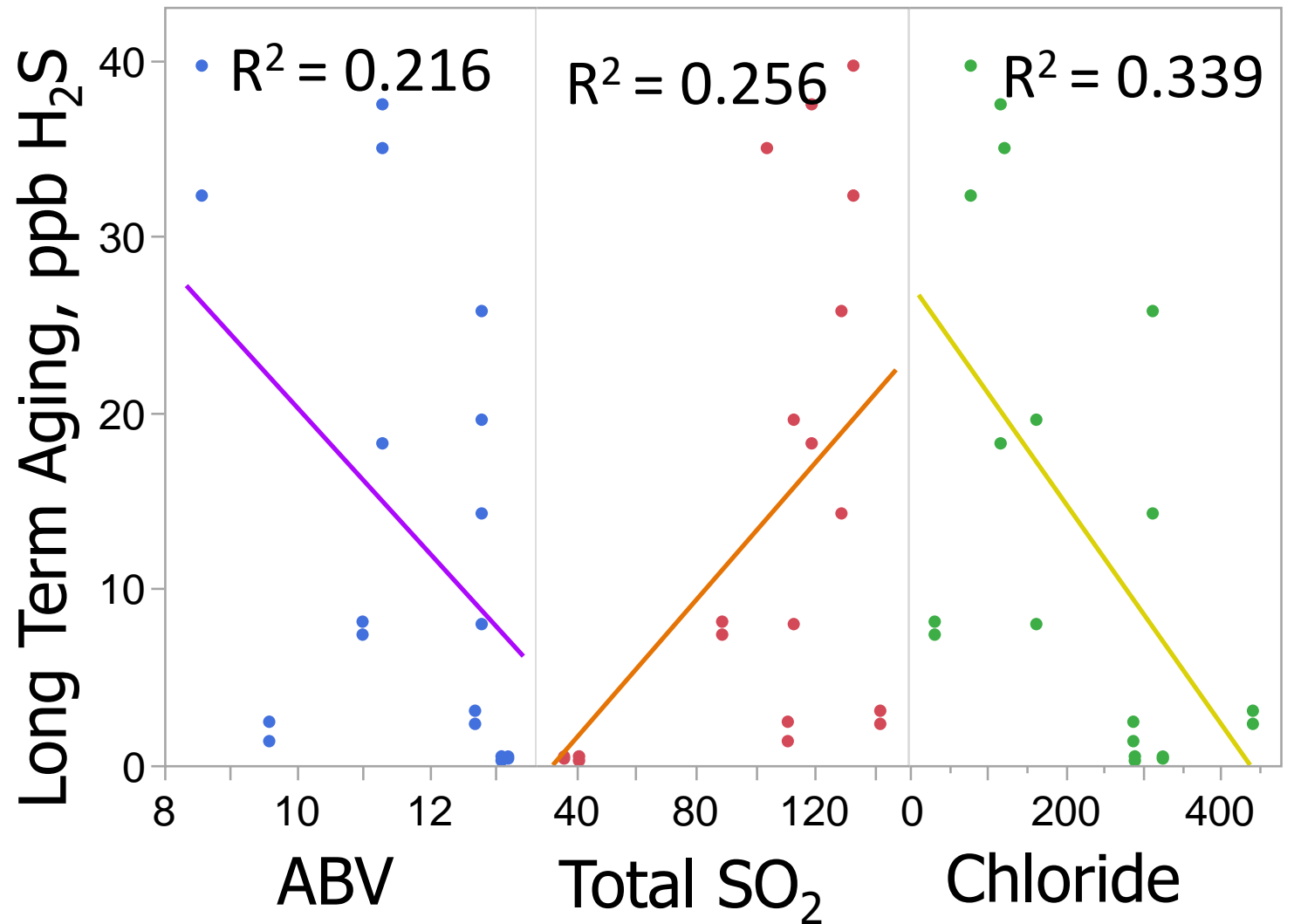
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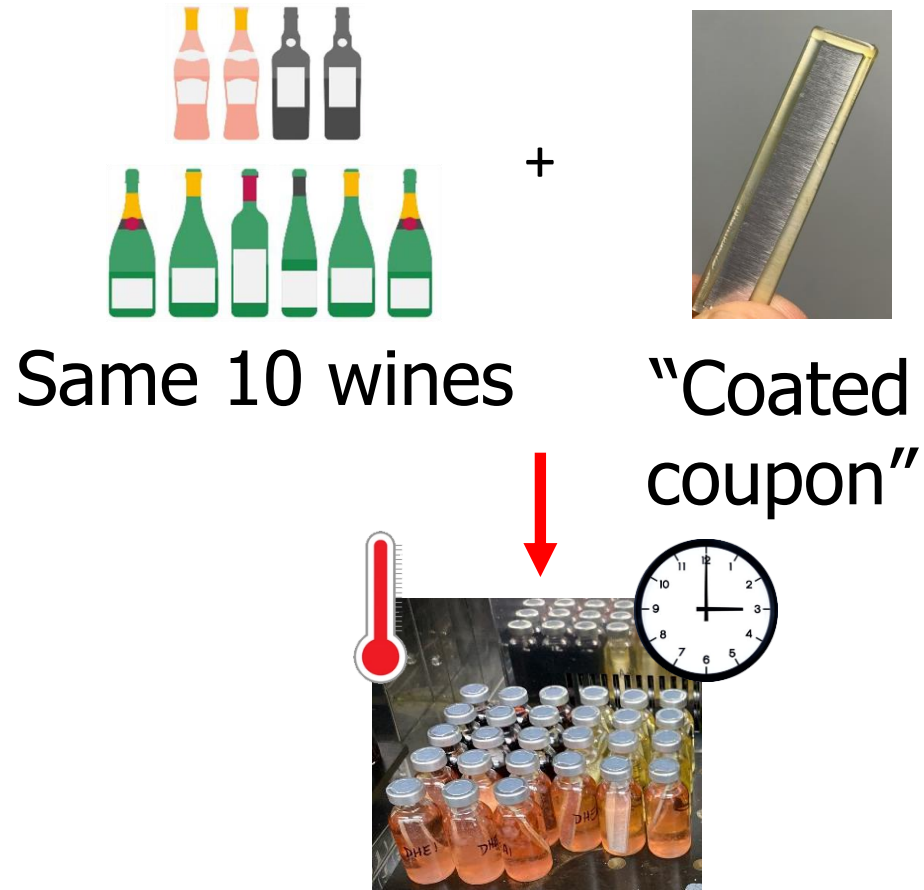
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Cu < 0.1 mg/L most samples

- Molecular SO₂ = best predictor of H₂S
- ~ 0.5 ppm target Molecular SO₂
- ABV, Total SO₂, Cl⁻ uncorrelated



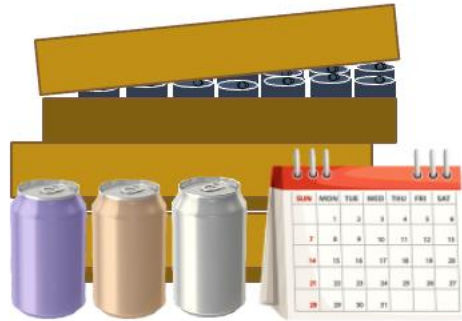
Can accelerated aging conditions predict H_2S formation?



Advantages

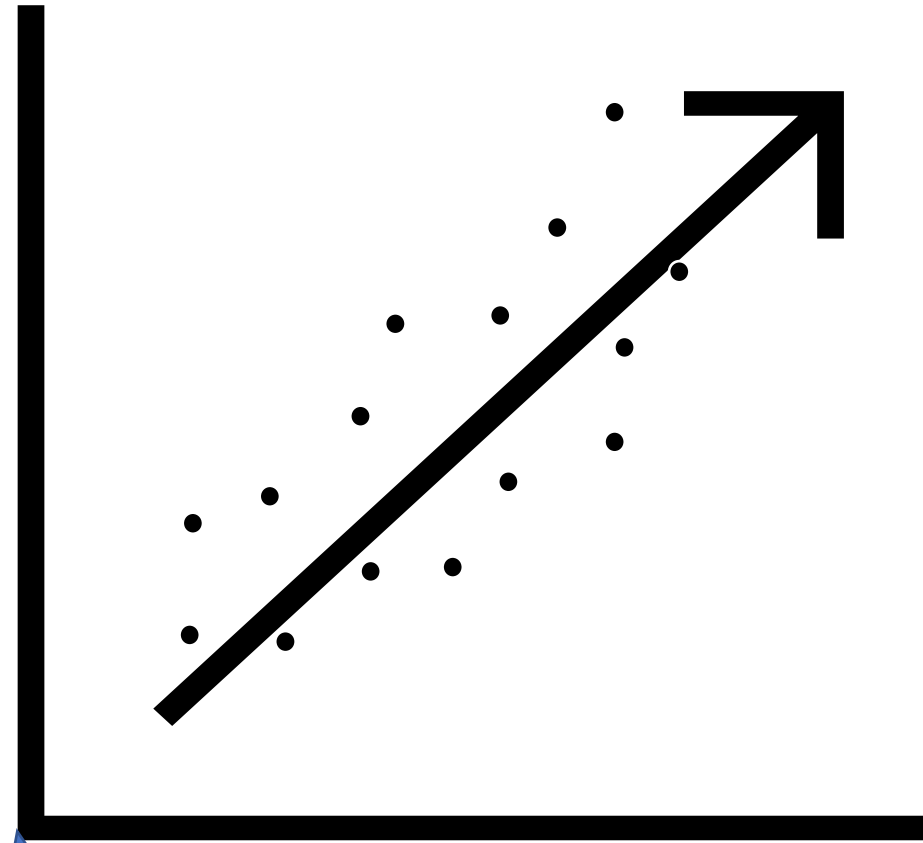
1. Increase sample throughput
2. Save \$\$ on cost of materials
3. Measure model systems easily

How can we validate accelerated aging?

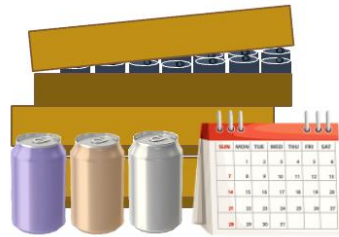


20 ° C, H₂S (ppb)

50 ° C, H₂S (ppb)

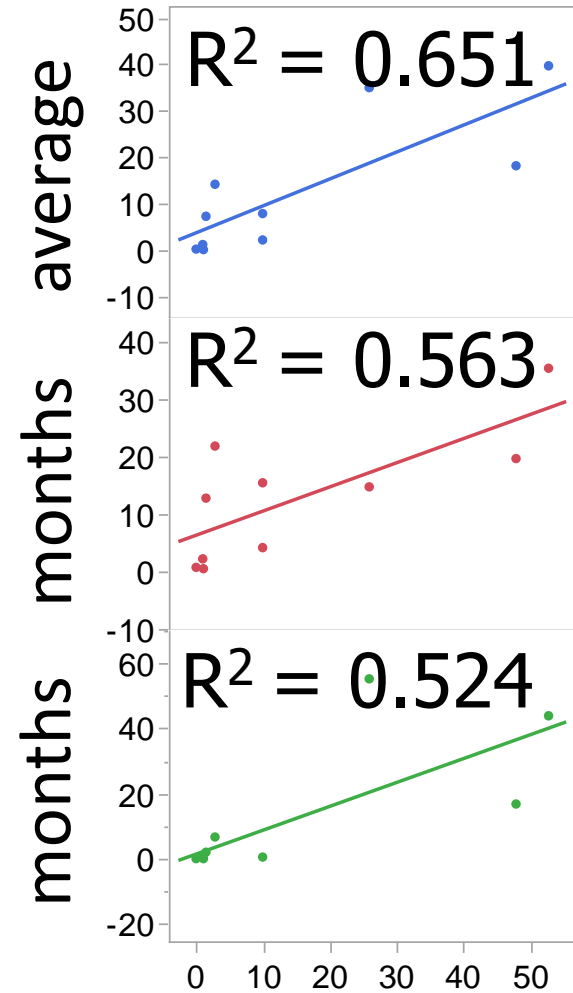


3 days 50 ° C predicts H₂S formation well



20 ° C, H₂S (ppb)

4 8 4/8 months average

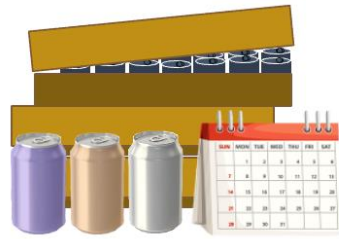


3 days

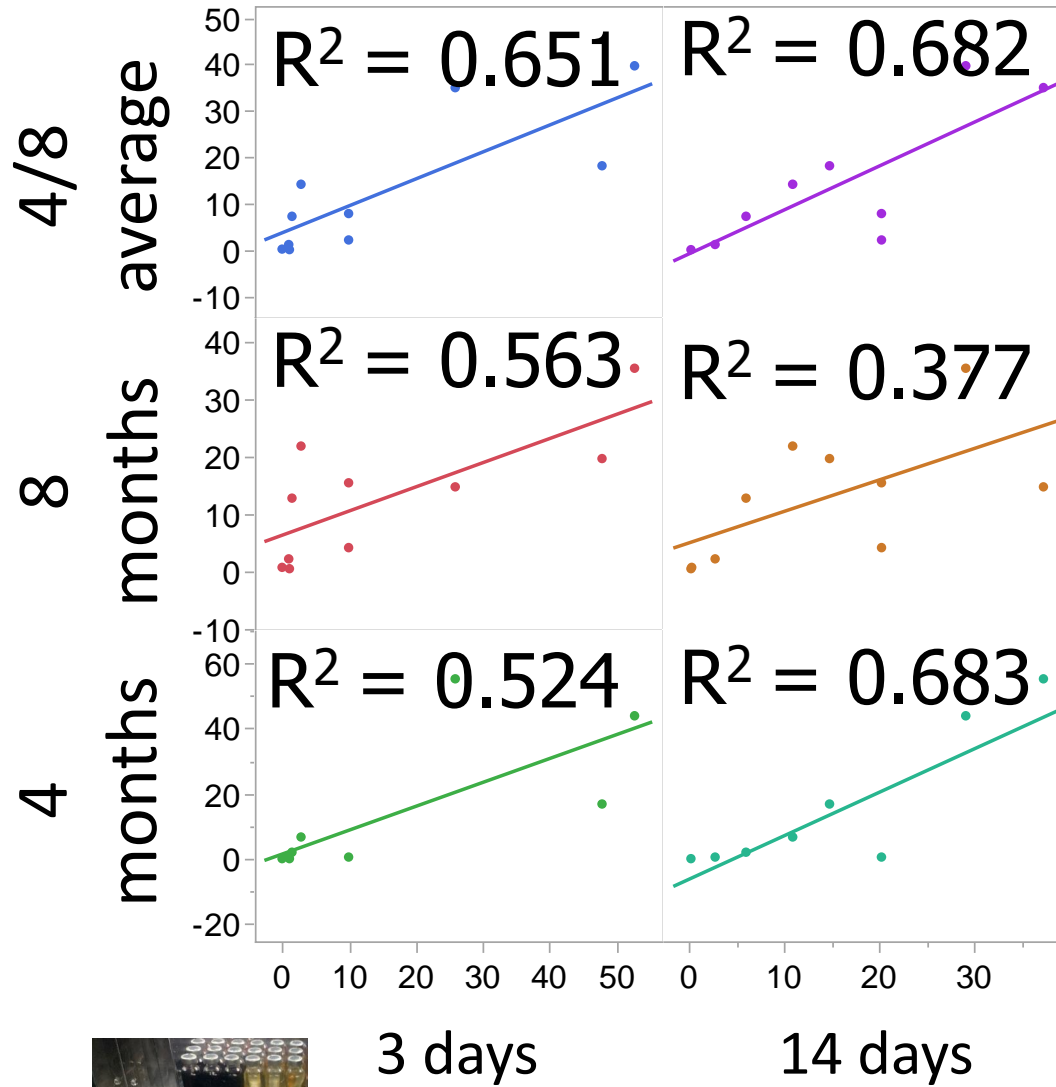


50 ° C, H₂S (ppb)

14 days also predicts H₂S formation well

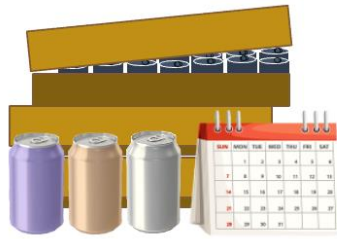


20 ° C, H₂S (ppb)

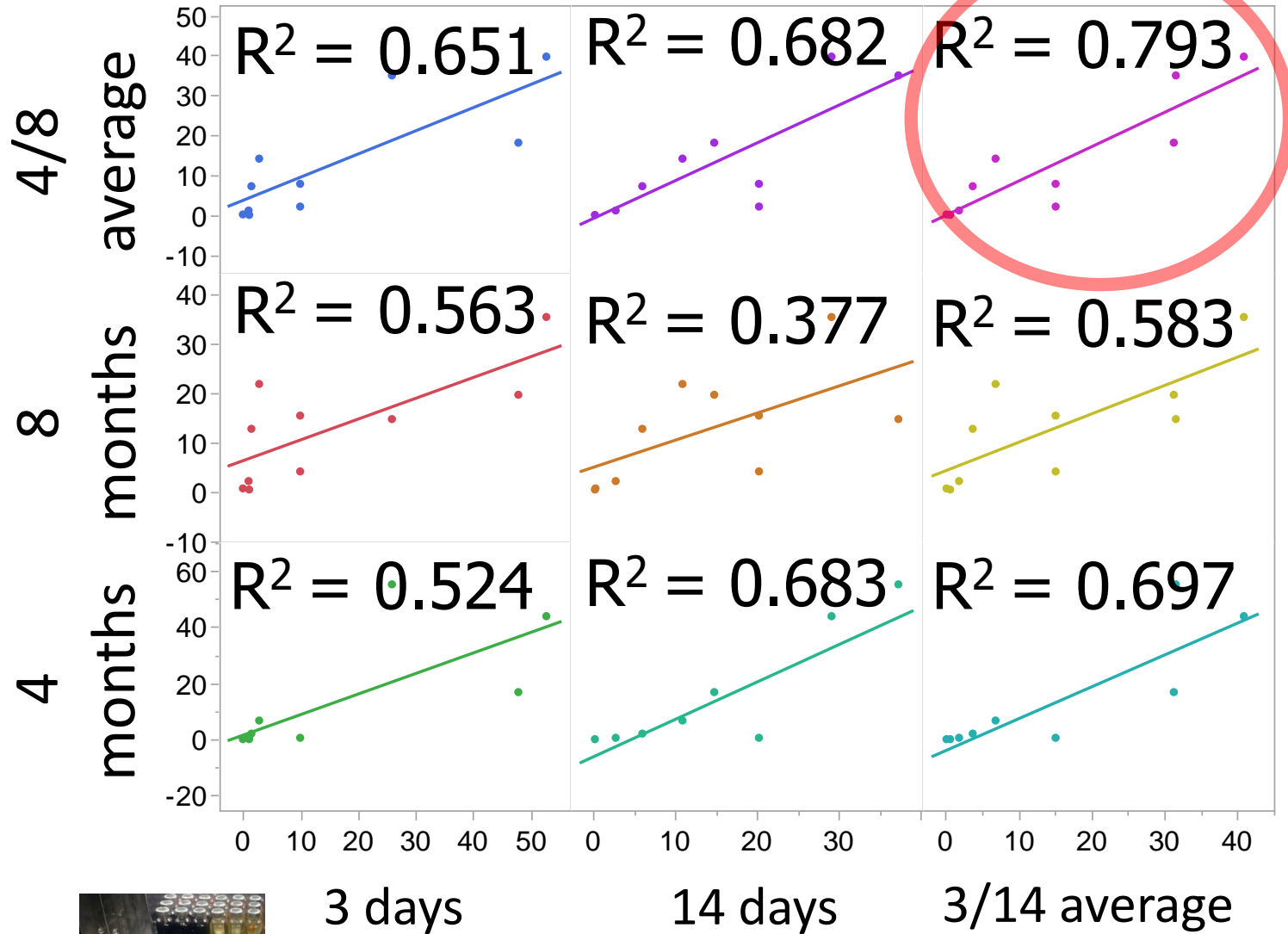


50 ° C, H₂S (ppb)

But, the average of 3 and 14 d predicts H₂S best



20 ° C, H₂S (ppb)



50 ° C, H₂S (ppb)

Putting accelerated aging test to work

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Now that we have a valid test,
we can use it to address
specific wine-systems!

Putting accelerated aging test to work: role of ethanol



Non-alcoholic wines
Red, Rosé, White



Ethanol, SO₂
Tartaric Acid



+ BPA-Epoxy
Coupon

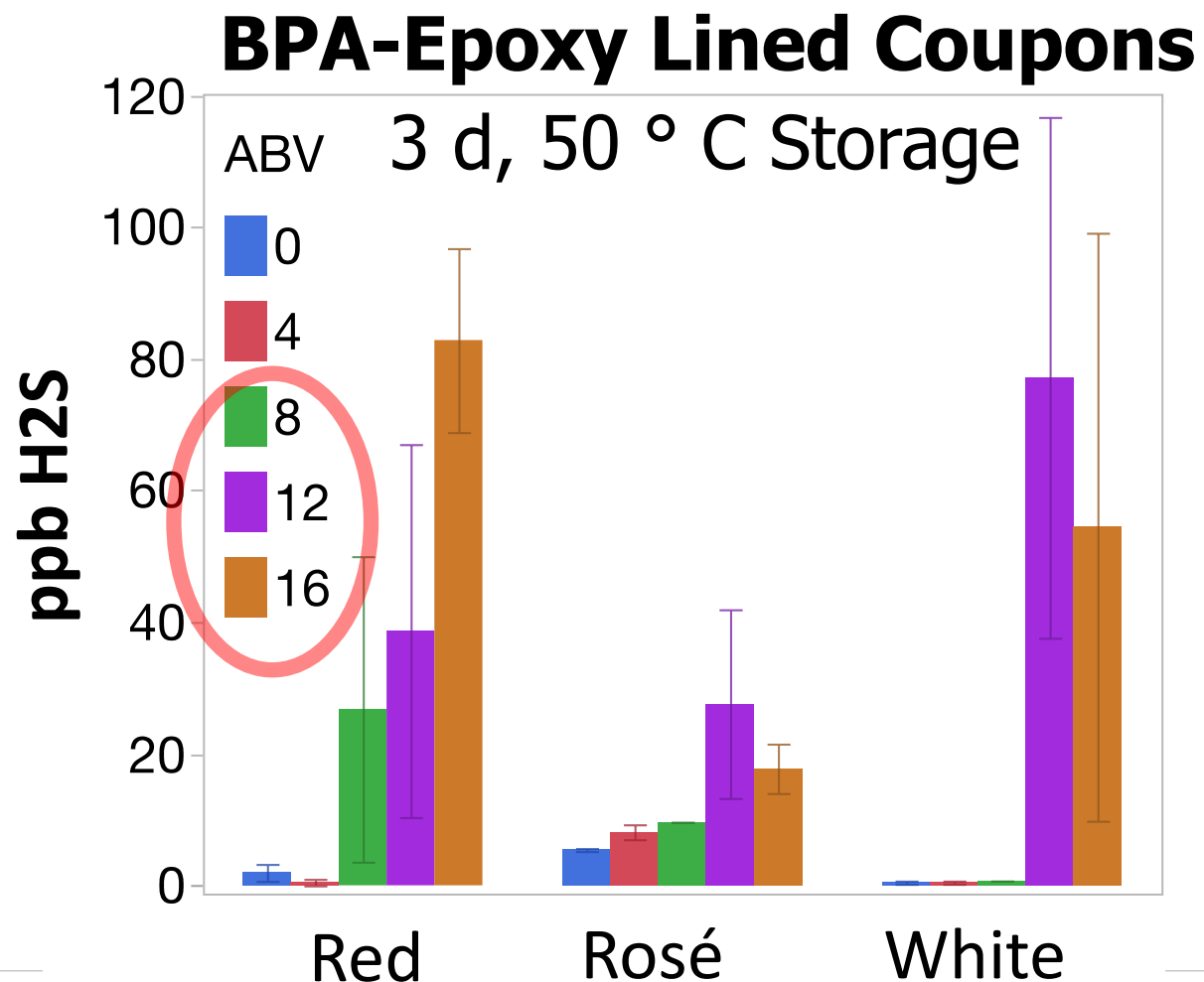


Challenge solutions with
specific wine chemistry

pH	3
Free SO ₂	50 ppm
EtOH	0, 4, 8, 12, 16 %

Putting accelerated aging test to work: role of ethanol

- ~8% ethanol “threshold” to observe H₂S production
- Ethanol shows a weak effect over the range of typical table wines

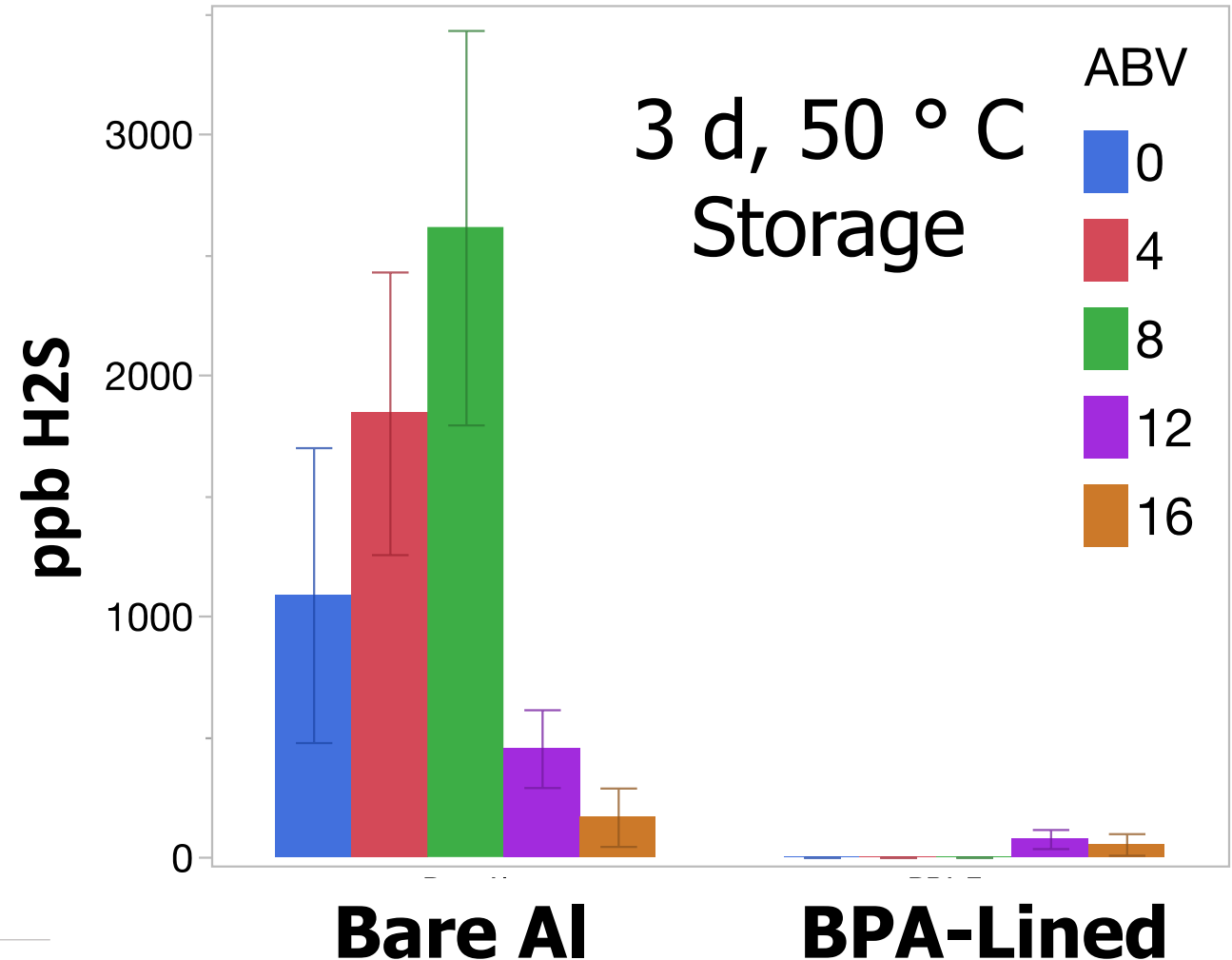


N = 3 for each bar, error bars are standard error.

Bare aluminum – Poor Predictor of H₂S

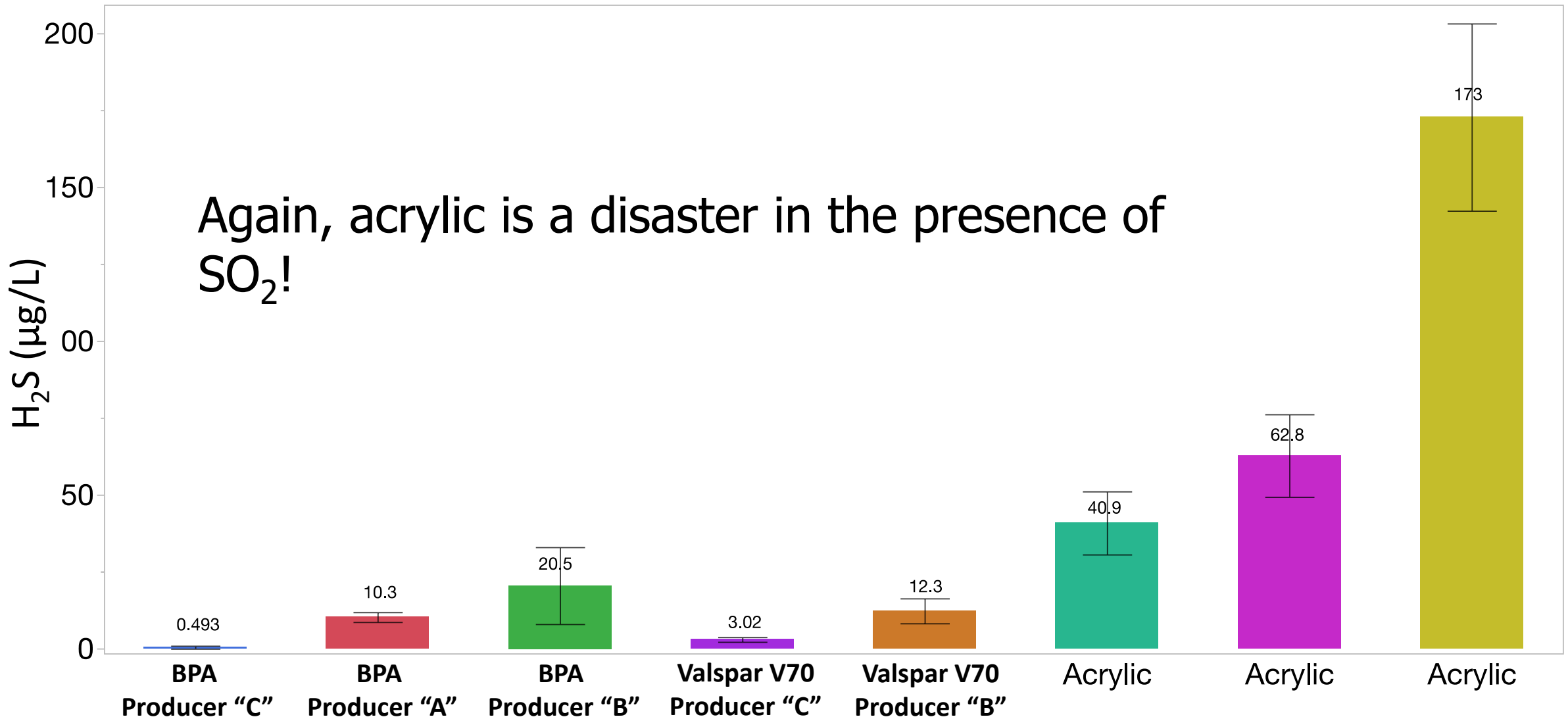
- Bare aluminum shows opposite ABV effect, with a peak at 8%.
- Don't use bare aluminum as a metric for H₂S potential!

White Wine, pH 3, 50 ppm Free SO₂



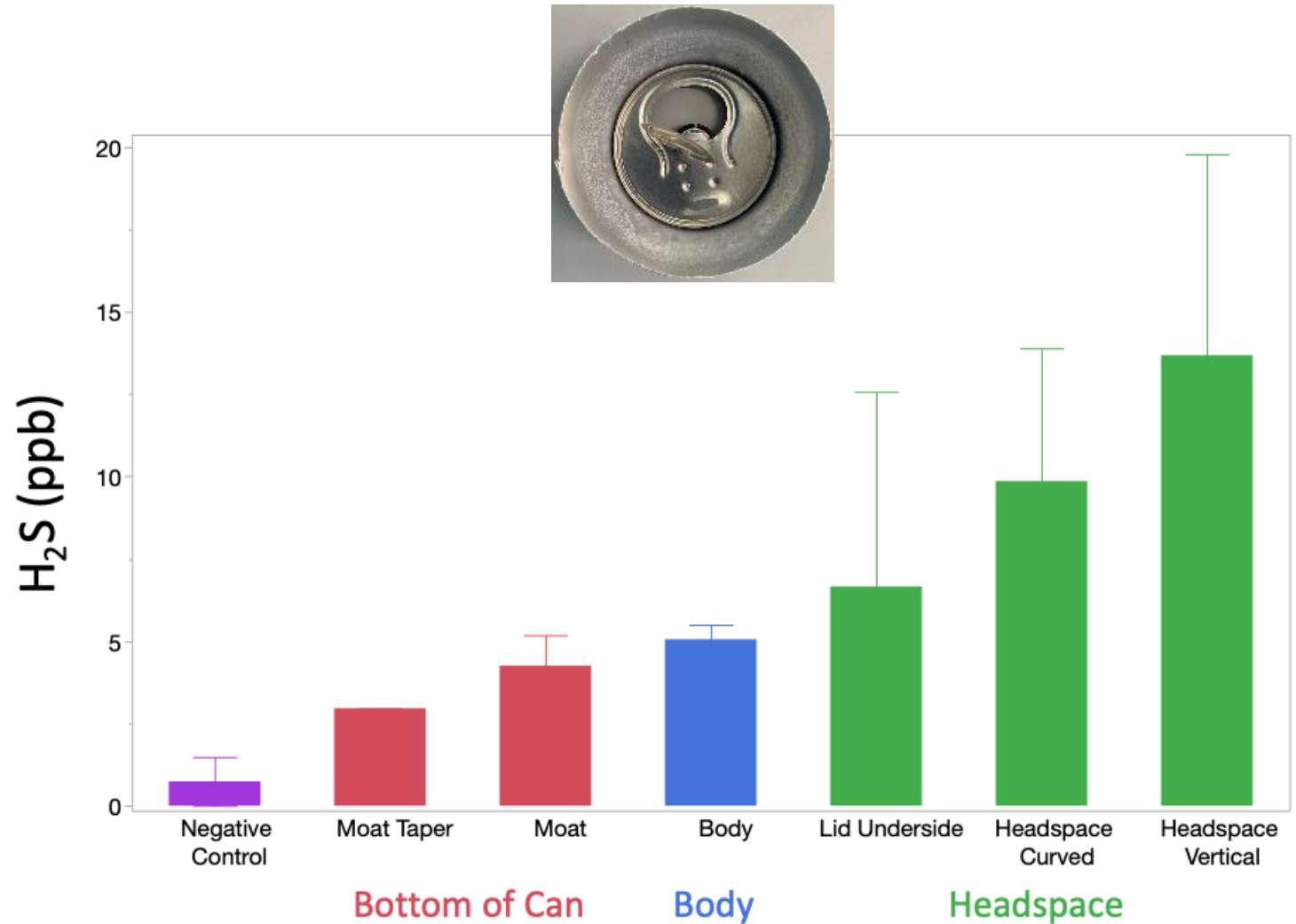
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Comparison of various liner performance



Accelerated tests allow us to isolate parts of the can

- Highest and most variable corrosion is in the headspace!
- Further suggests that neutral, volatile acids are correlated with corrosion.



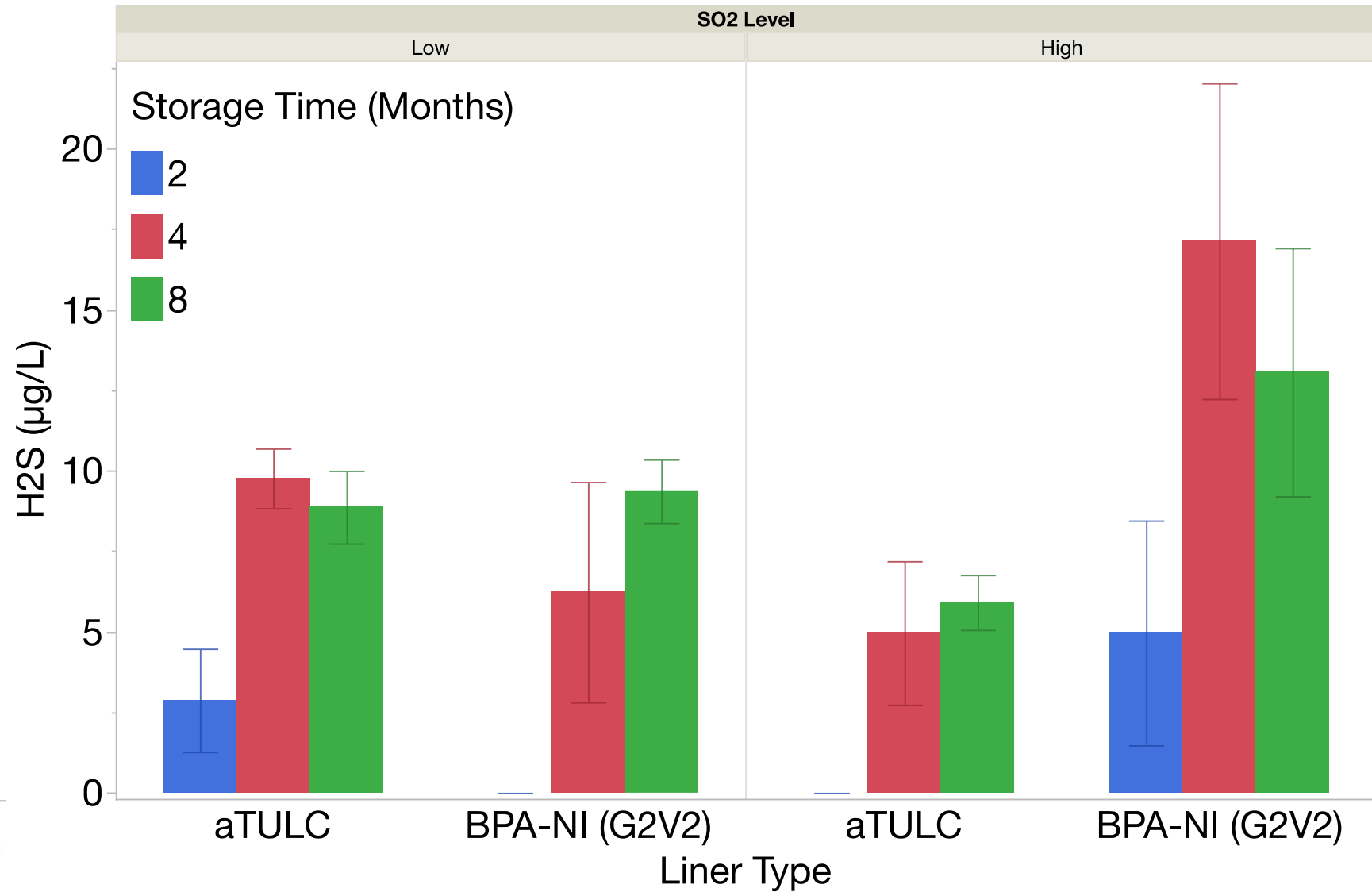
aTULC cans are a promising new liner



- aTULC cans are formed from extra thick PET-coated aluminum
 - More consistent application compared to spray on coatings.
 - Internal data (American Canning) suggests minimal scalping.

Preliminary data suggests aTULC is good

- aTULC performed better than G2V2 at 4 and 8 months for a high SO₂ wine.
- Need more trials to confirm effect.



For Wines / Ciders

- Keep molecular SO_2 under 0.5 mg/L.
 - If high residual sugar, in-bulk HPP and velcorin are options.
- Try to stay away from acrylic liners whenever possible – BPA and BPA-NI are comparable in performance.
 - Get in contact w/ me if interested about accelerated aging protocol to test.
- Preliminary results suggest that aTULC is “good”!
 - More trials need to be conducted

General Takeaways

- When conducting shelf-life tests, the brand and type of liner matters!
 - Bare aluminum should not be used under any circumstance.
 - Producer A liner \neq Producer B liner even if same polymer
- Be sure to use as many replicates as possible – these systems are highly variable... yet one bad sample can ruin a whole pallet!
- We don't fully understand this system! We are currently working on finding new metrics that can predict corrosion more completely than simply Al^{3+} or H_2S .

Acknowledgements



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- ASBC and Master Brewers
- New York State Wine and Grape Foundation

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